

Electrical measurements at the nanoscale using Scanning Probe Microscopy - Part 2 -

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Question

**Can we probe the dielectric constant
(electric permittivity)
on the nanoscale?**

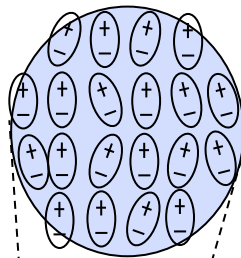
Basics of electric polarization

Macroscopic polarization

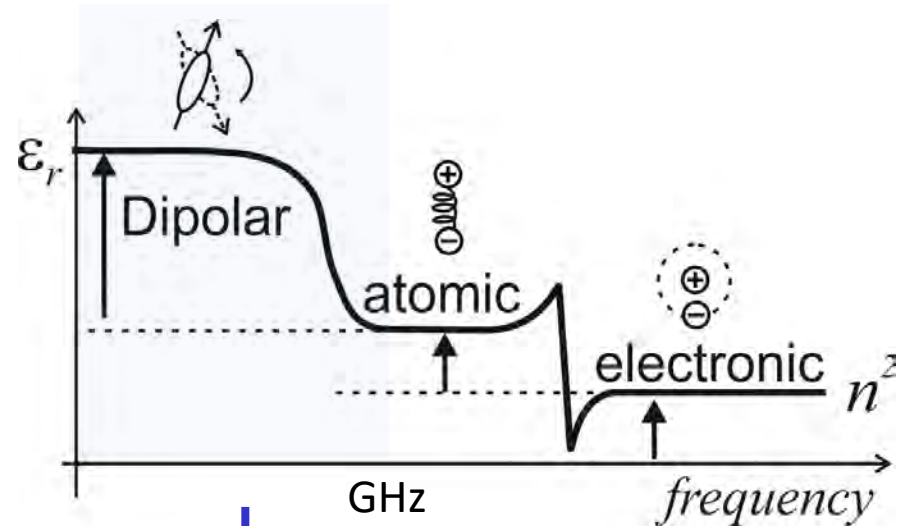
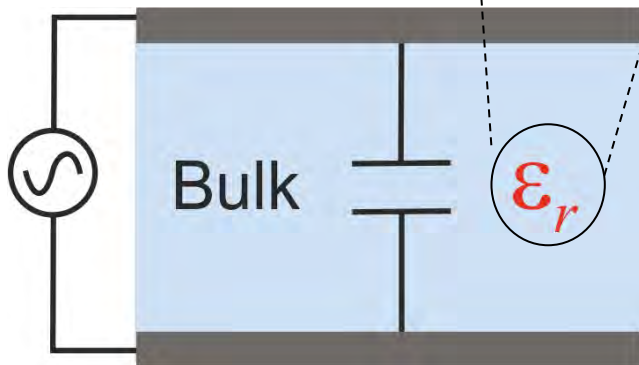
$$\vec{P}(\vec{r}) = \sum_i \vec{p}_i = \epsilon_0 (\epsilon_r - 1) \vec{E}(\vec{r})$$

Dielectric constant
or permittivity

Electric dipoles $\vec{p}_i$
distribution, order,
rotational freedom



E



Capacitance measurement

$$C = \frac{\epsilon_0 \epsilon_r A}{h} \quad \begin{array}{l} A = \text{area} \\ h = \text{thickness} \end{array}$$

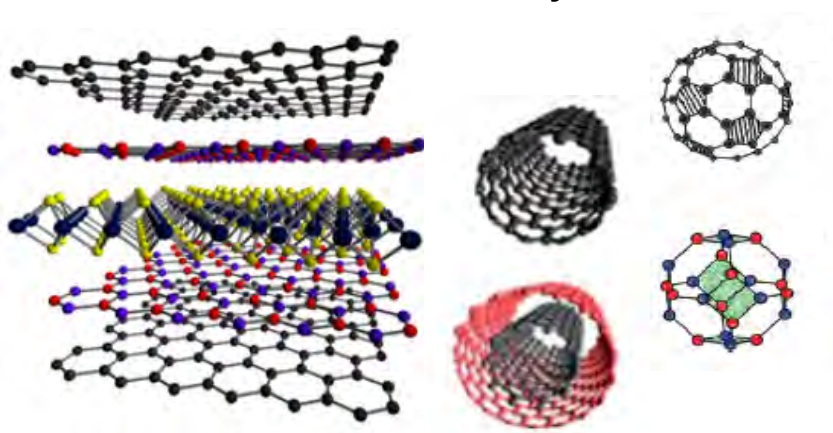
Unknown dielectric constants at the nanoscale

Lack of experimental tools with sufficient sensitivity

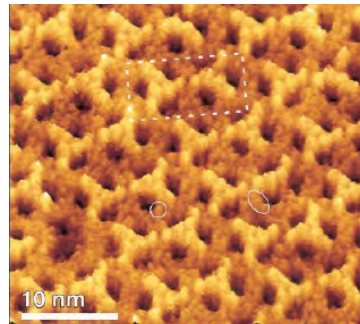
Theory to be established

A multitude of systems still to explore

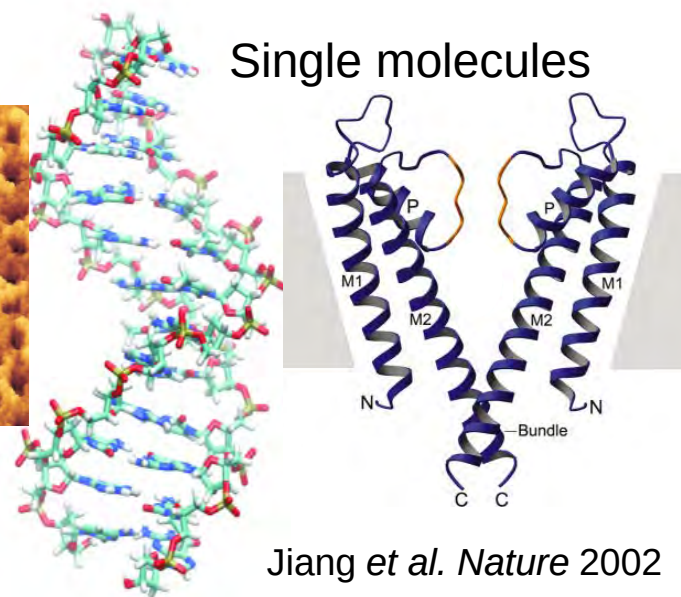
Low-dimensional systems



Biointerfaces



Single molecules



Bonaccorso *et al.* ACS Nano 2013

Goldberg *et al.* ACS Nano 2010

Jiang *et al.* Nature 2002

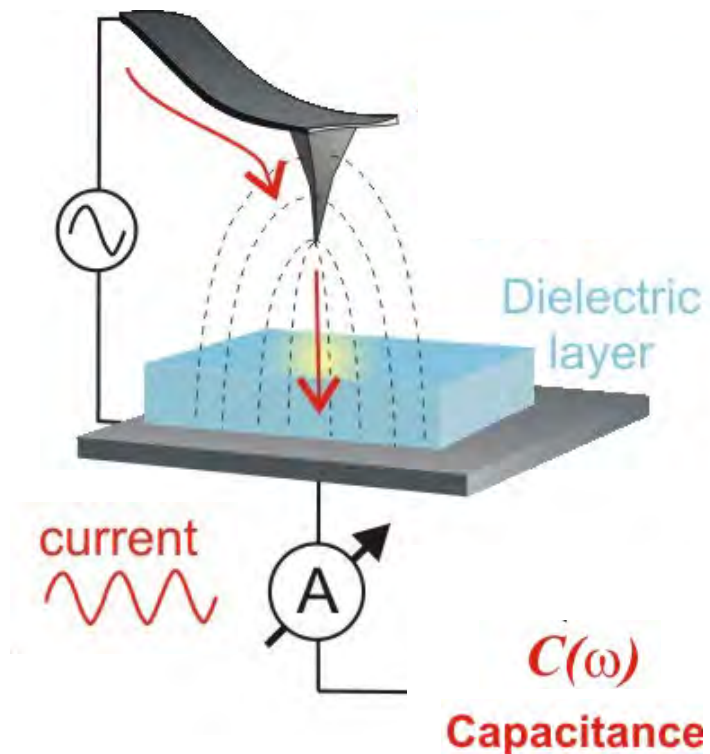
Engel *et al.* Nat. Struct. Biol. 2000

Physical Sciences

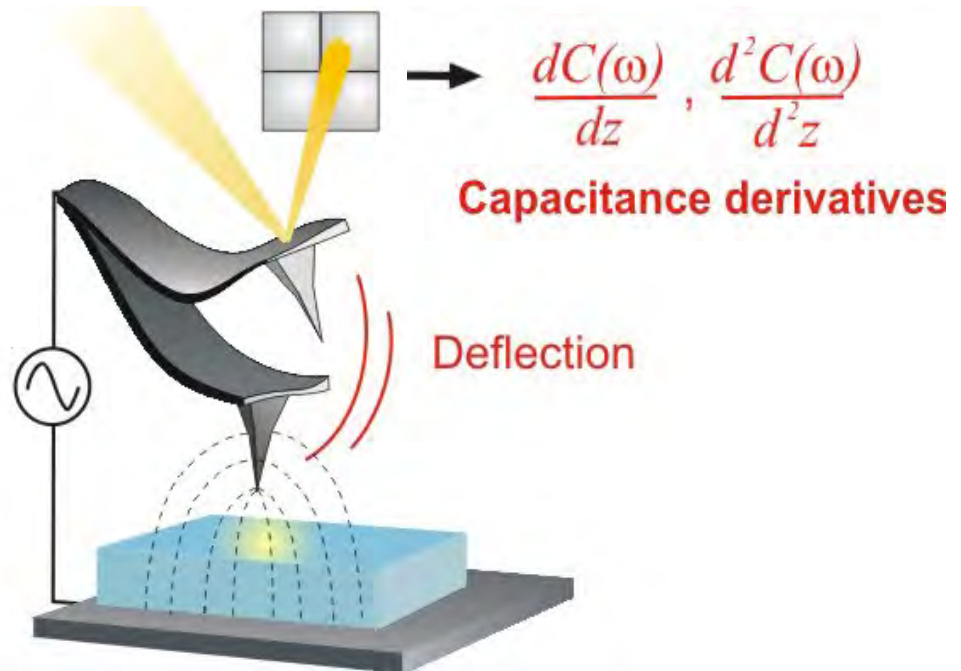
Chemistry and Biology

Scanning Dielectric Microscopy (SDM)

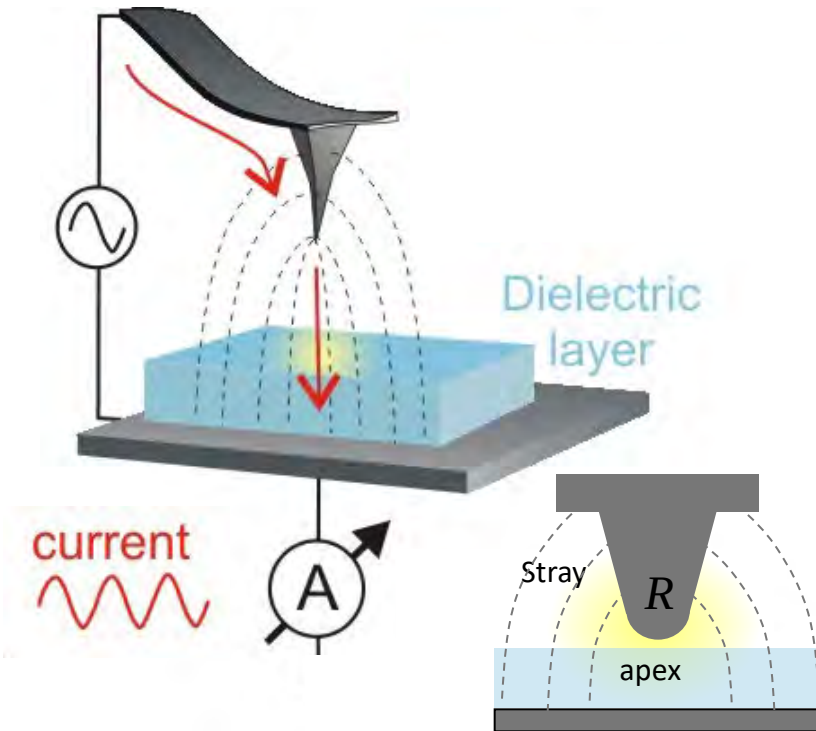
Current-sensing



Electrostatic force-sensing

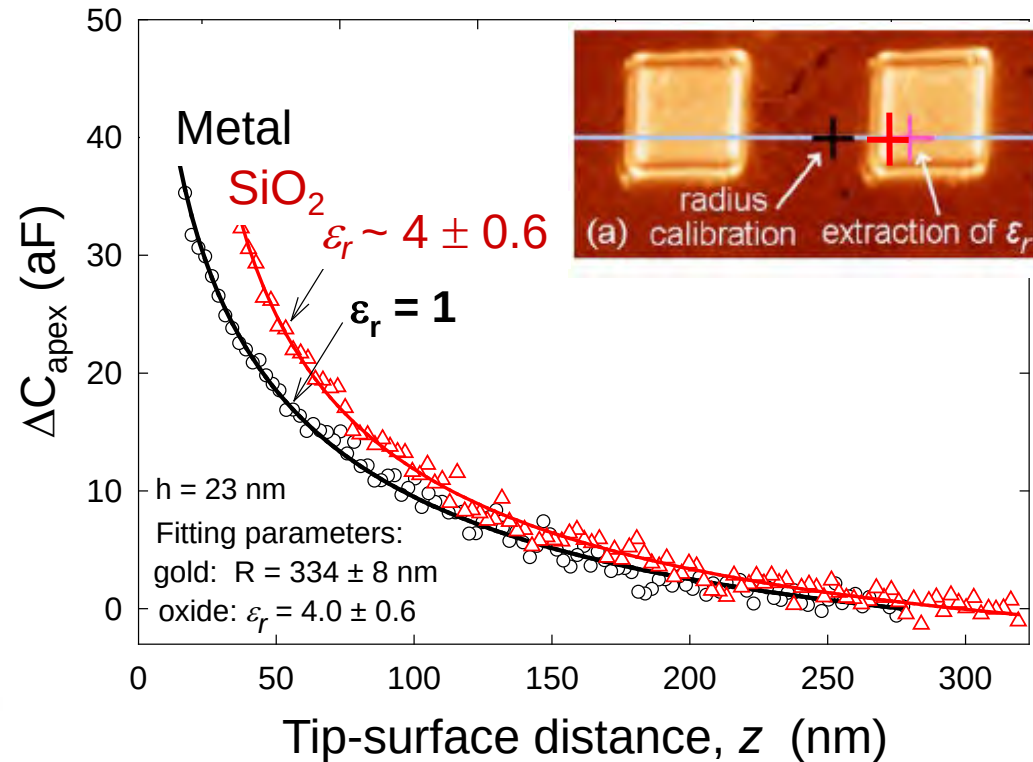


Dielectric constant measurement with CS-AFM



$$C_{total} = C_{apex} + C_{cone} + C_{cantilever} + C_{setup}$$

$$\Delta C_T \approx \Delta C_{apex} + k_{stray}(z_0 - z)$$



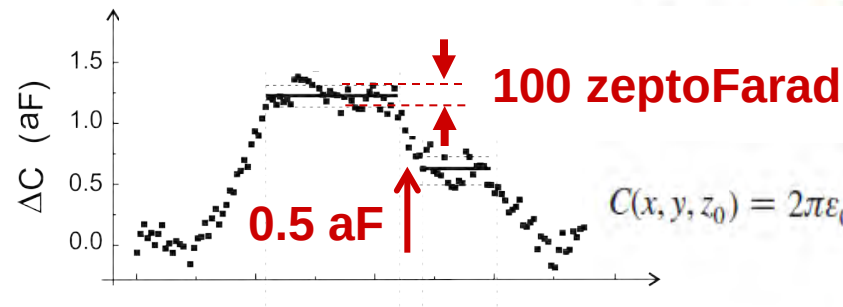
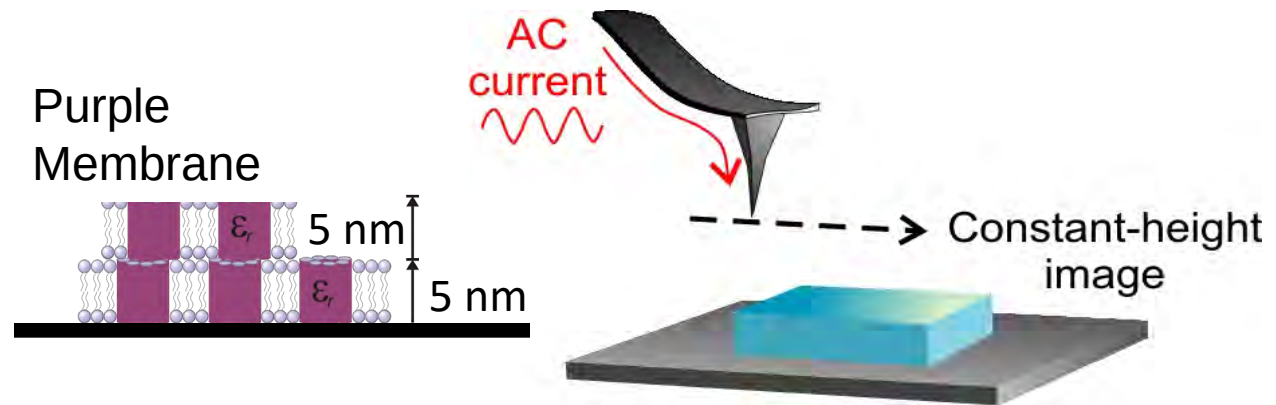
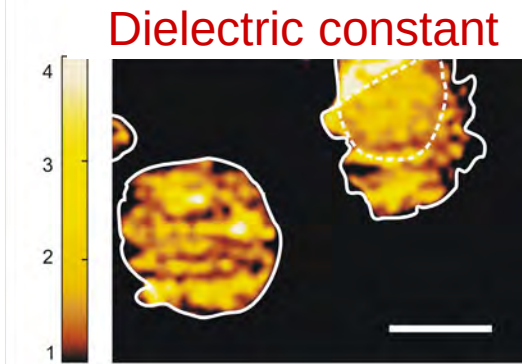
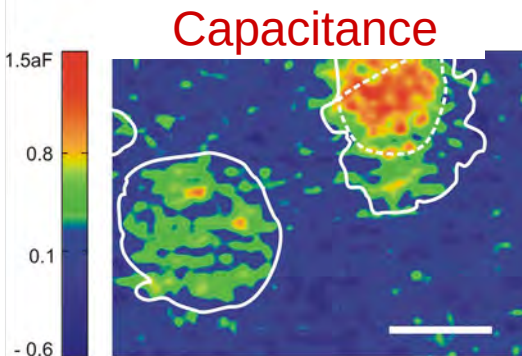
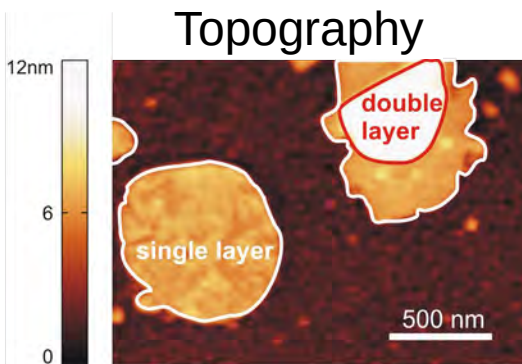
For thin dielectrics $h < 50$ nm

$$C_{apex}(z, h, R, \epsilon_r) = 2\pi\epsilon_0 R \cdot \ln \left(1 + \frac{R \cdot (1 - \sin \theta)}{z + \frac{h}{\epsilon_r}} \right) + c_0$$

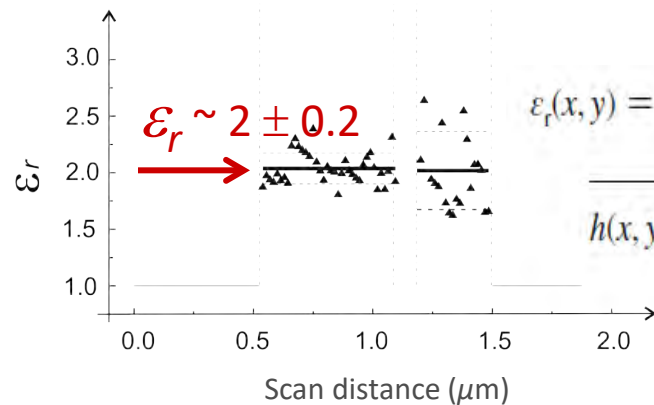
Logarithmically dependent on ϵ

Proportional to the tip radius R

Dielectric constant imaging with CS-AFM



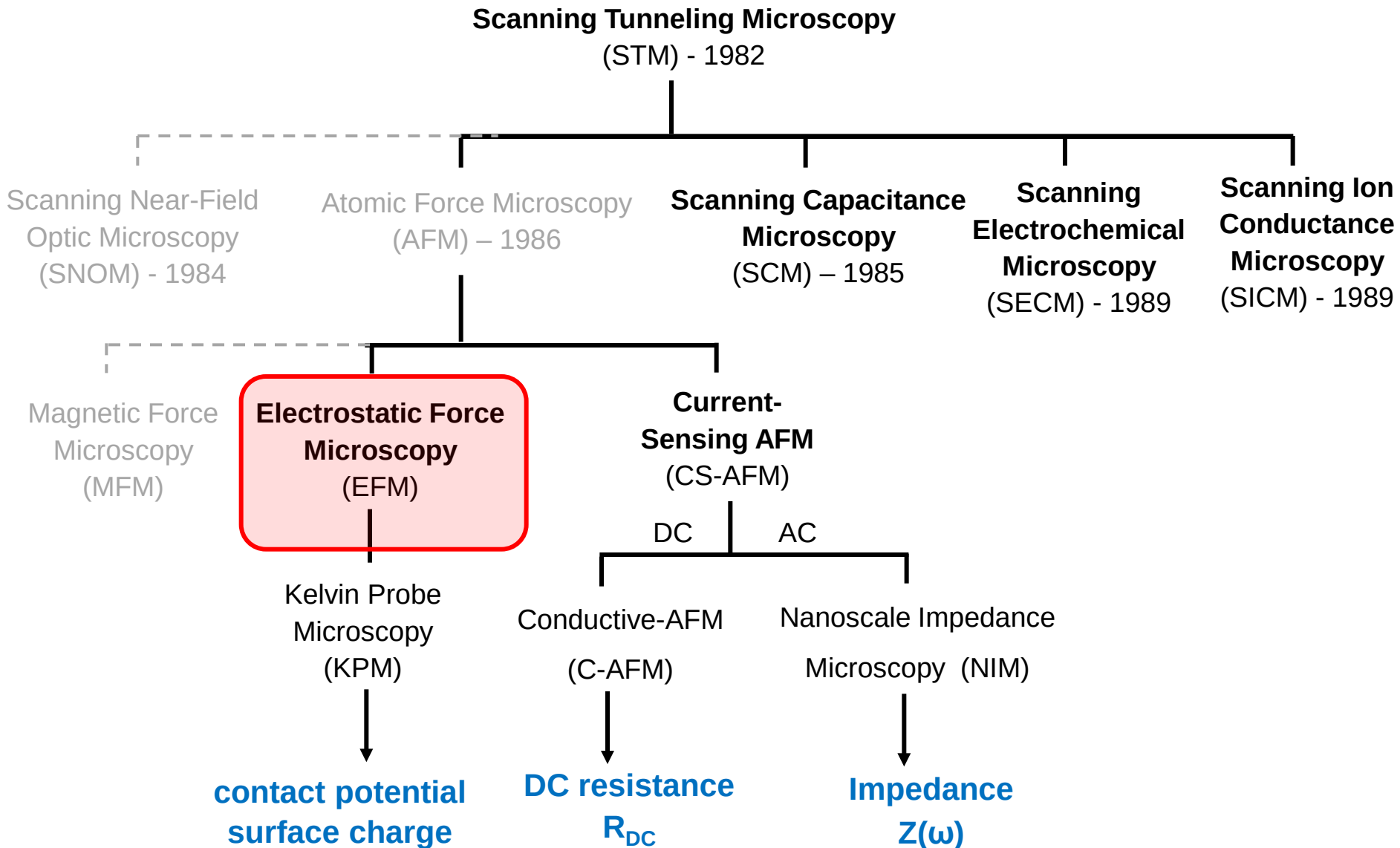
$$C(x, y, z_0) = 2\pi\epsilon_0 R \ln \left[1 + \frac{R(1 - \sin \vartheta)}{z_0 + \frac{h(x, y)(1 - \epsilon_r(x, y))}{\epsilon_r(x, y)}} \right]$$



$$\epsilon_r(x, y) =$$

$$\frac{h(x, y)}{h(x, y) - z_0 + \frac{R(1 - \sin \vartheta)}{\exp\left(\frac{\Delta C(x, y, z_0)}{2\pi\epsilon_0 R}\right)\left(1 + \frac{R(1 - \sin \vartheta)}{z_0}\right) - 1}}$$

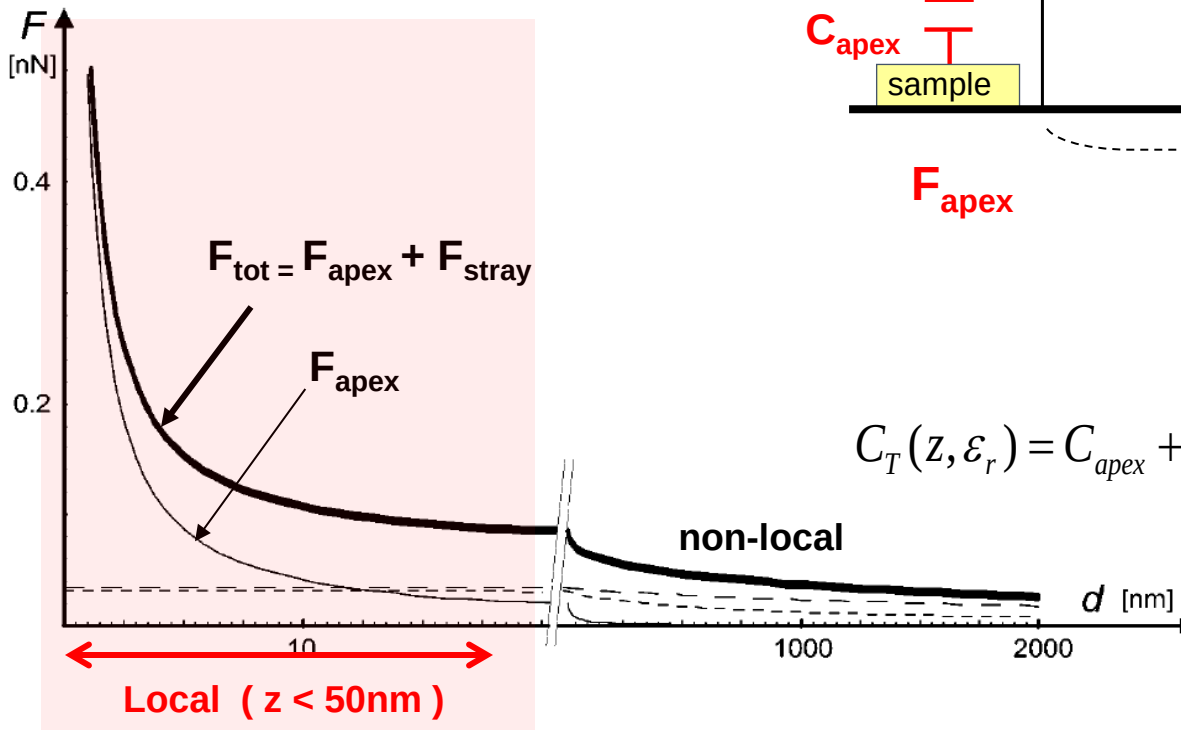
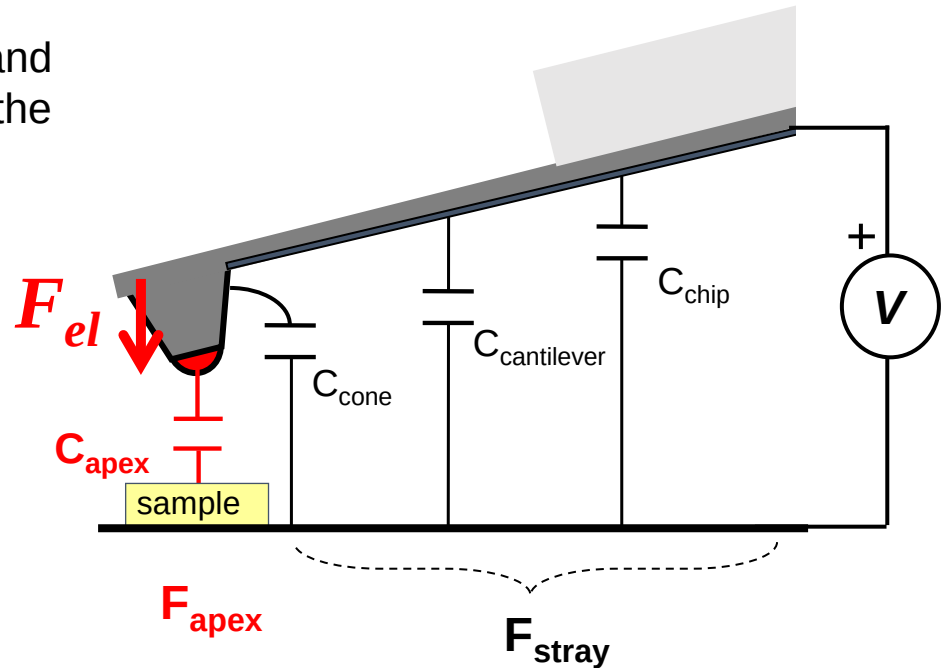
Electrical Scanning Probe Microscopies



Electrostatic Force Microscopy (EFM)

When a voltage is applied between the tip and the cantilever kept **out of contact**, the cantilever detects an electrostatic force

$$F_{el}(z) = -\frac{1}{2} \frac{\partial C}{\partial z} V^2$$



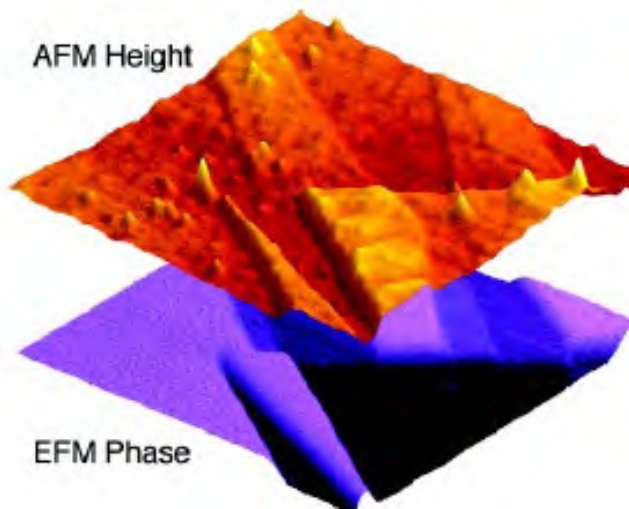
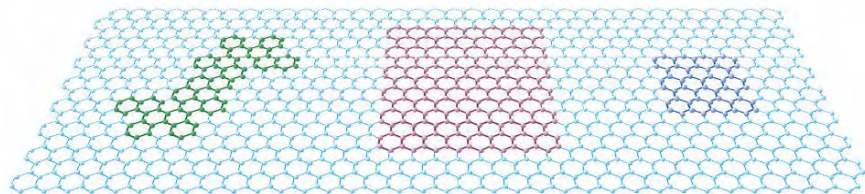
$$C_T(z, \epsilon_r) = C_{apex} + C_{cone} + C_{cantilever} + C_{chip}$$

The cantilever is negligible (pointing to $C_{cantilever}$)

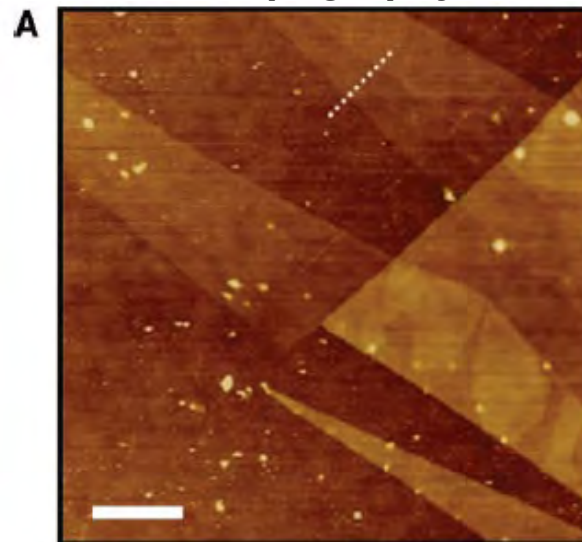
The chip not detected (pointing to C_{chip})

Electrostatic Force Microscopy (EFM)

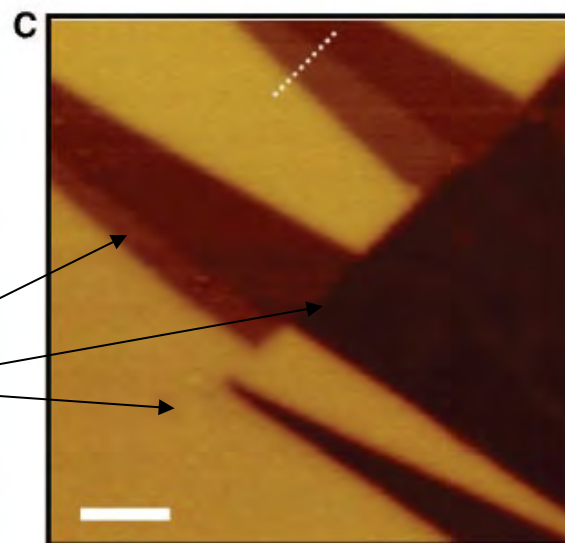
Graphene
flat monolayer of carbon atoms
packed into a 2D sheet



Topography

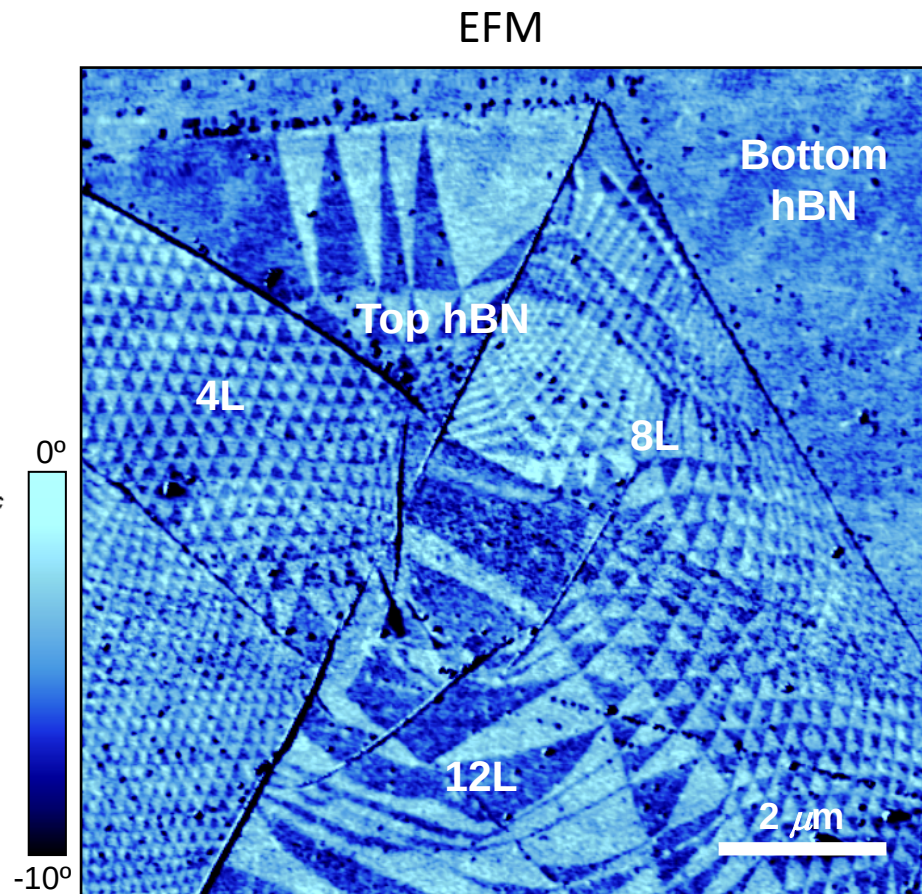
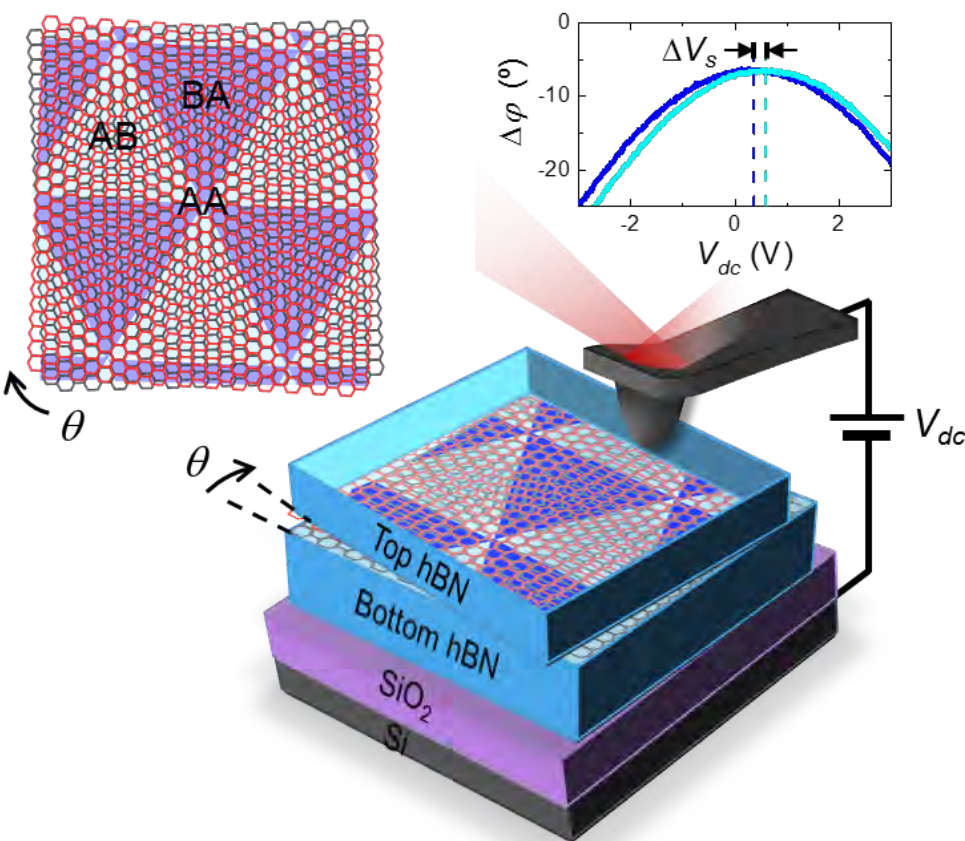


EFM



Ferroelectric-like polarization in twisted hBN

Emergence of interfacial dipolar domains

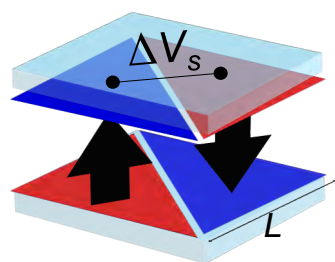
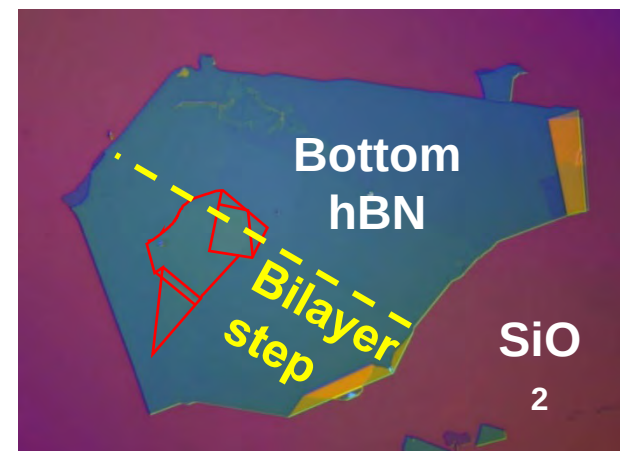
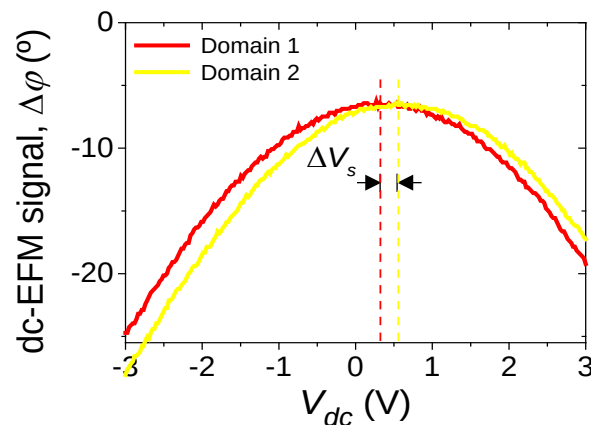
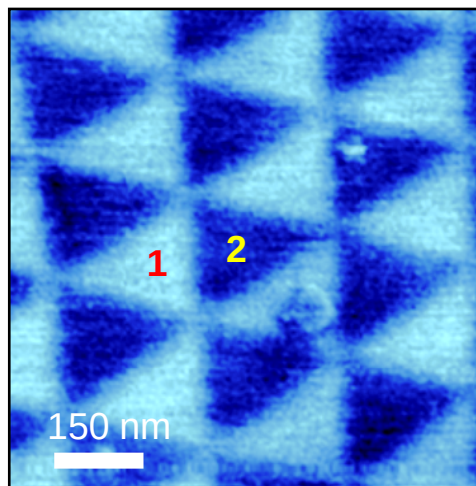


Oppositely charged domains
due to BN and NB interfacial dipoles
when twisting two insulating crystals

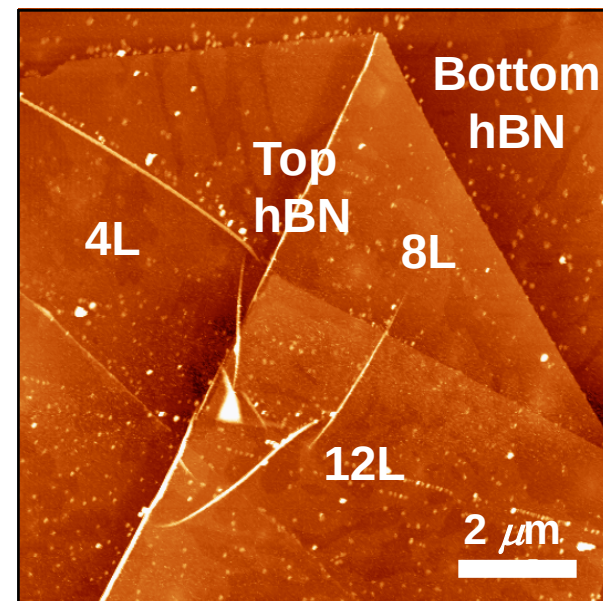
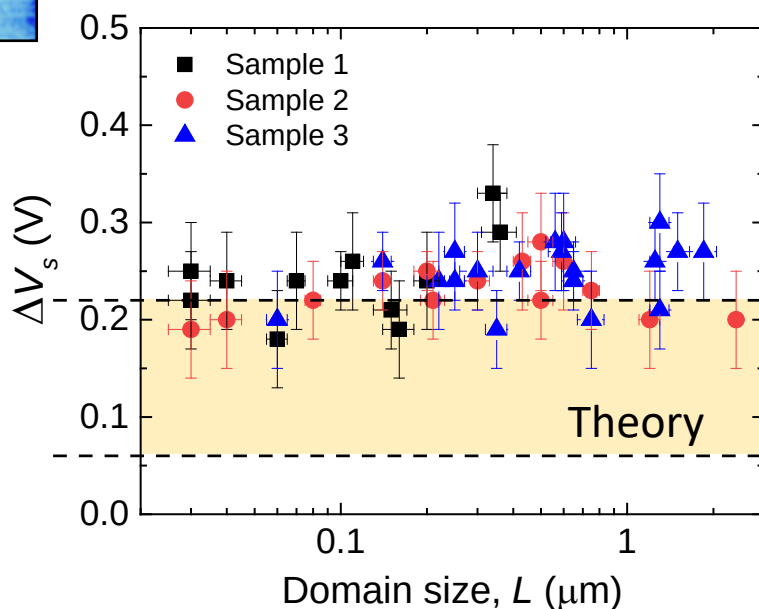
Ferroelectric-like polarization in twisted hBN

Emergence of interfacial dipolar domains

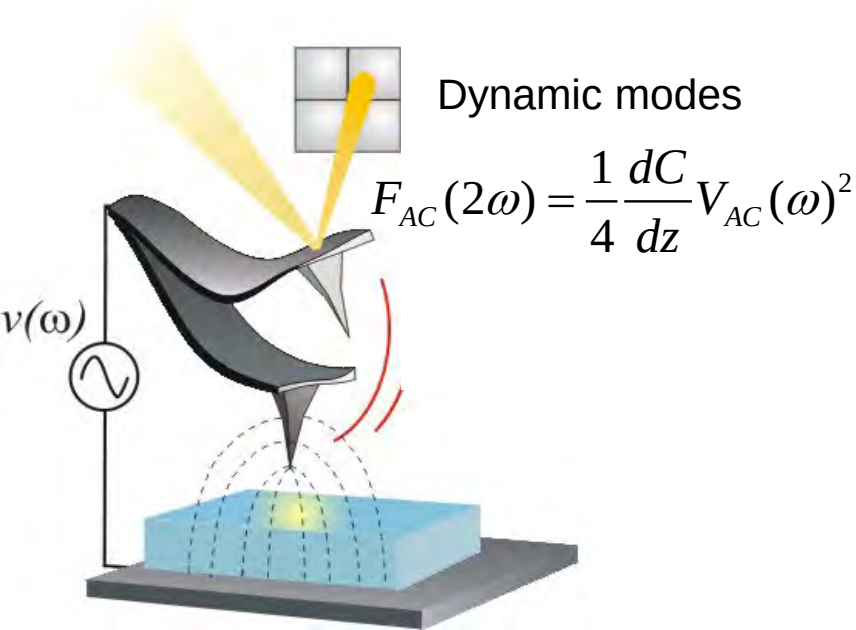
EFM



BN and NB
interfacial dipoles



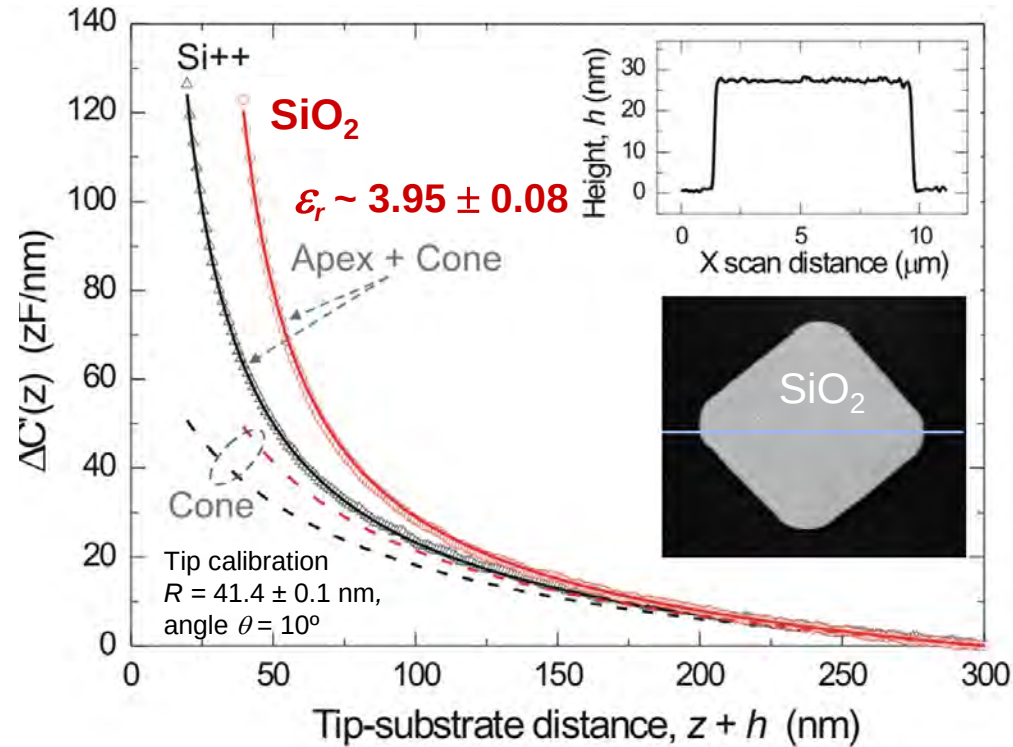
Dielectric constant measurement using EFM



negligible

$$C_{total} = C_{apex} + C_{cone} + \overset{\text{negligible}}{C_{cantilever}} + \cancel{C_{setup}}$$

$$\frac{dC_{total}}{dz}(z, R, h, \epsilon_r) \approx \frac{dC_{apex}}{dz} + \frac{dC_{cone}}{dz}$$



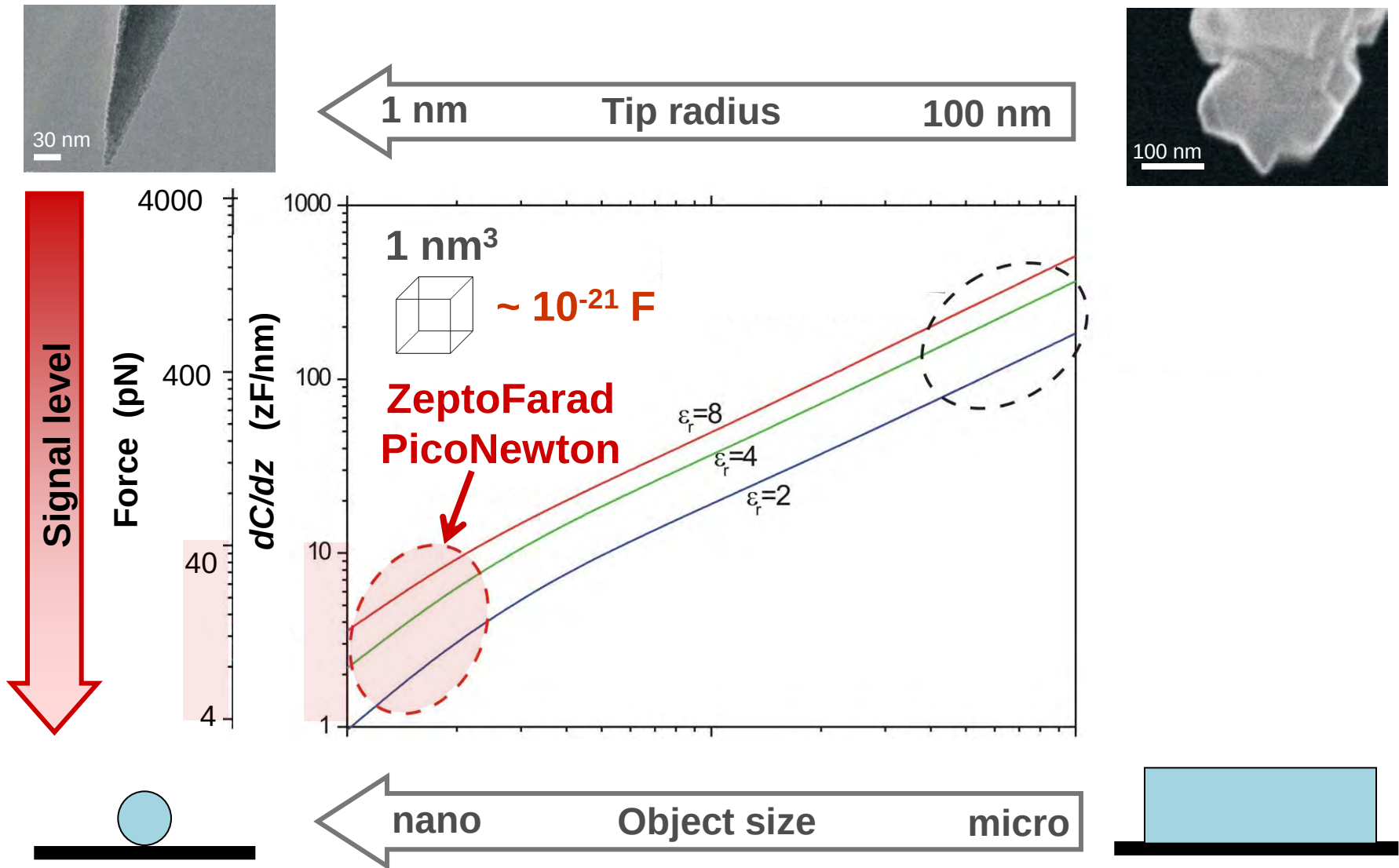
Advantages

Higher sensitivity, less parasitics

Disadvantages

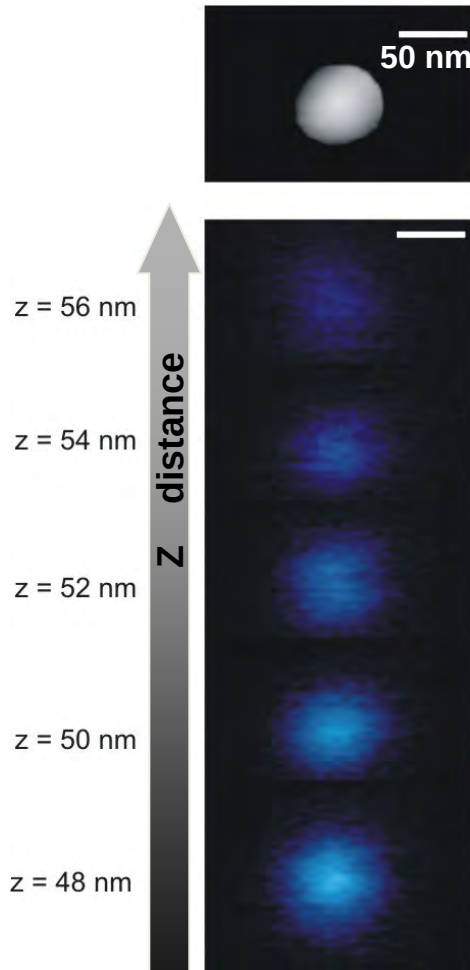
**Cantilever mechanical resonances,
cantilever bandwidth**

Ultra-high sensitivity in capacitance is needed

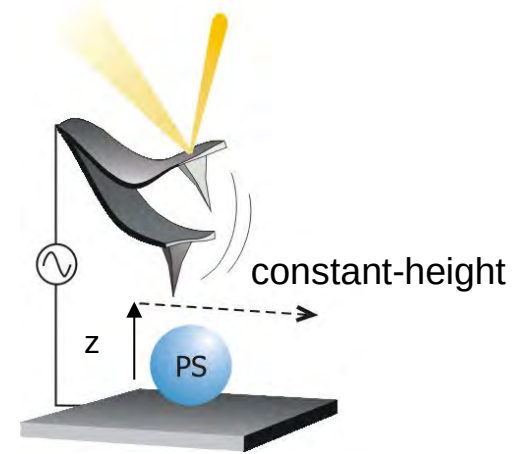
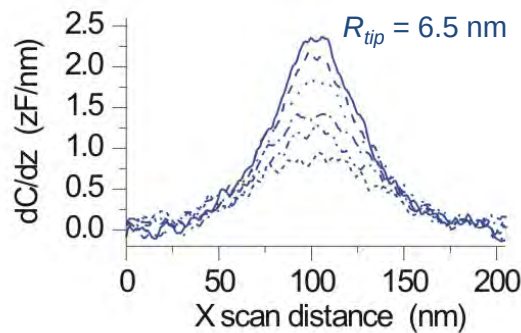
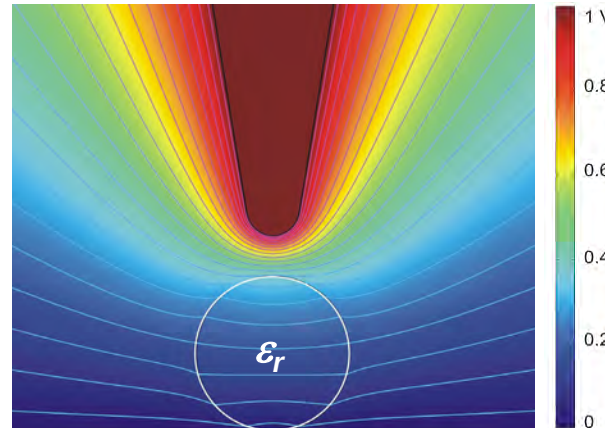


Dielectric constant of a single nanoparticle

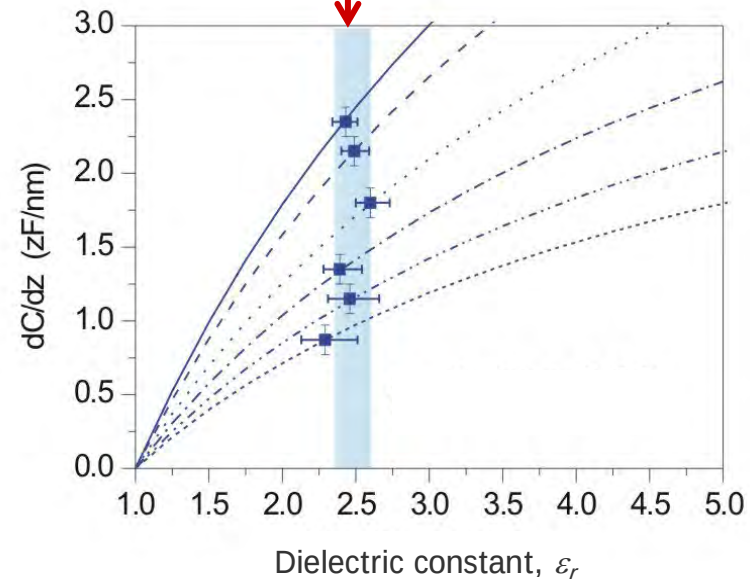
Polystyrene ($\epsilon_r = 2.6$)



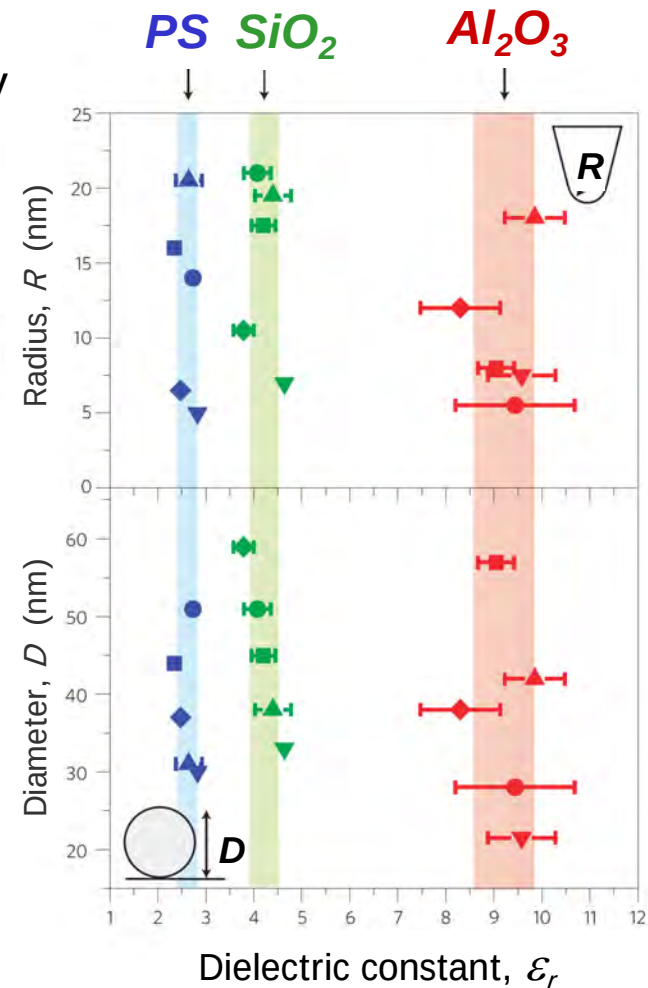
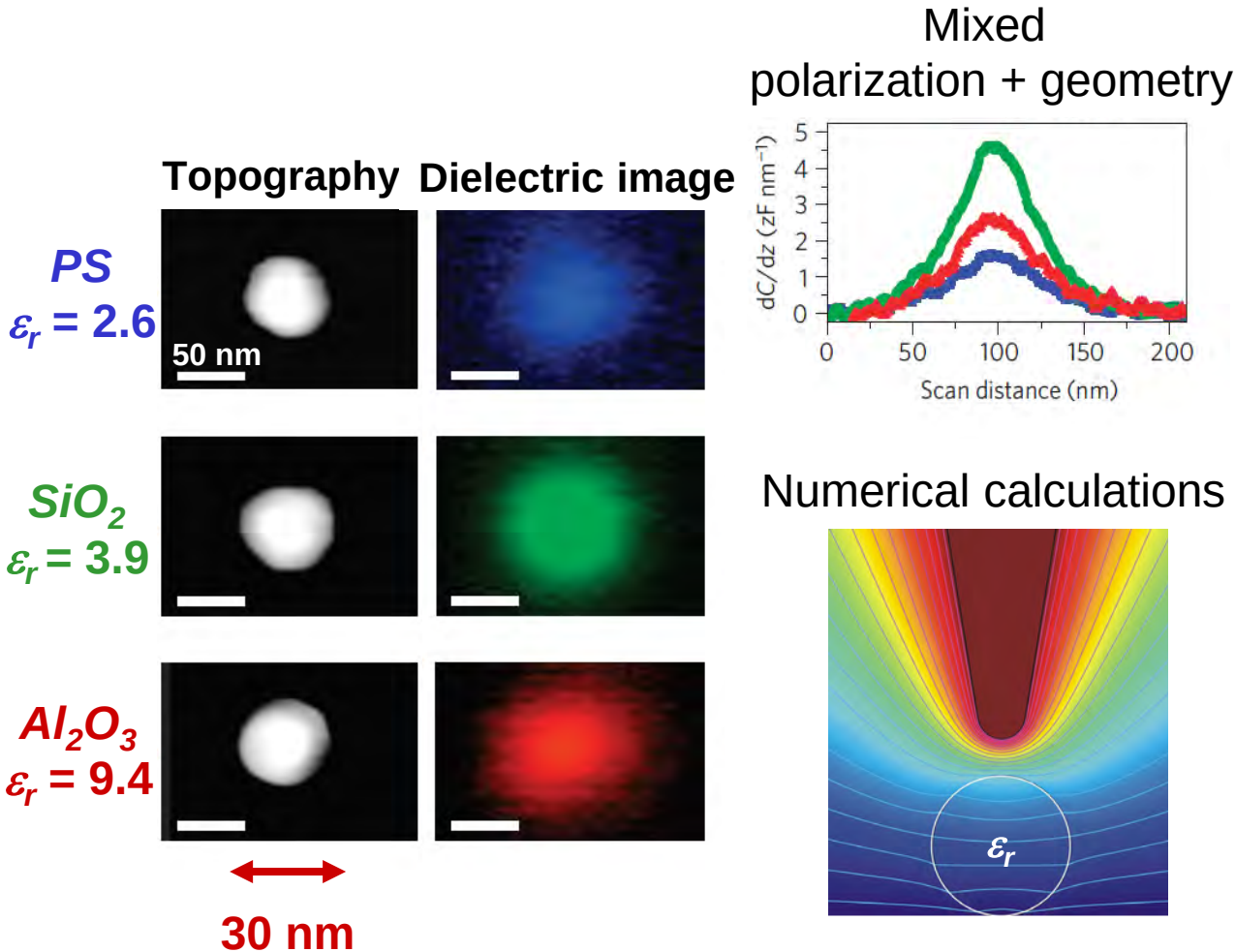
Numerical calculations



$$\epsilon_r \sim 2.44 \pm 0.09$$



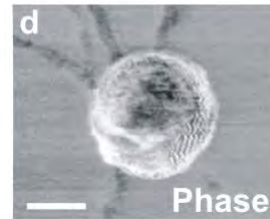
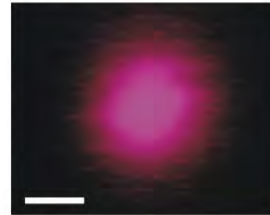
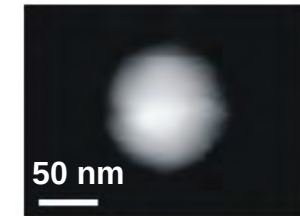
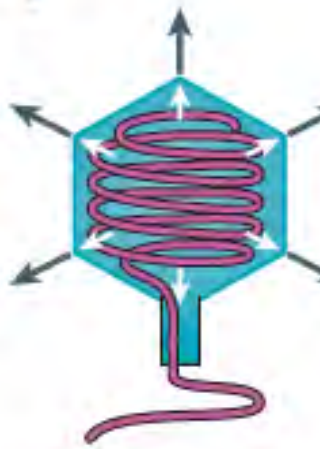
Dielectric constant of single nanoparticles



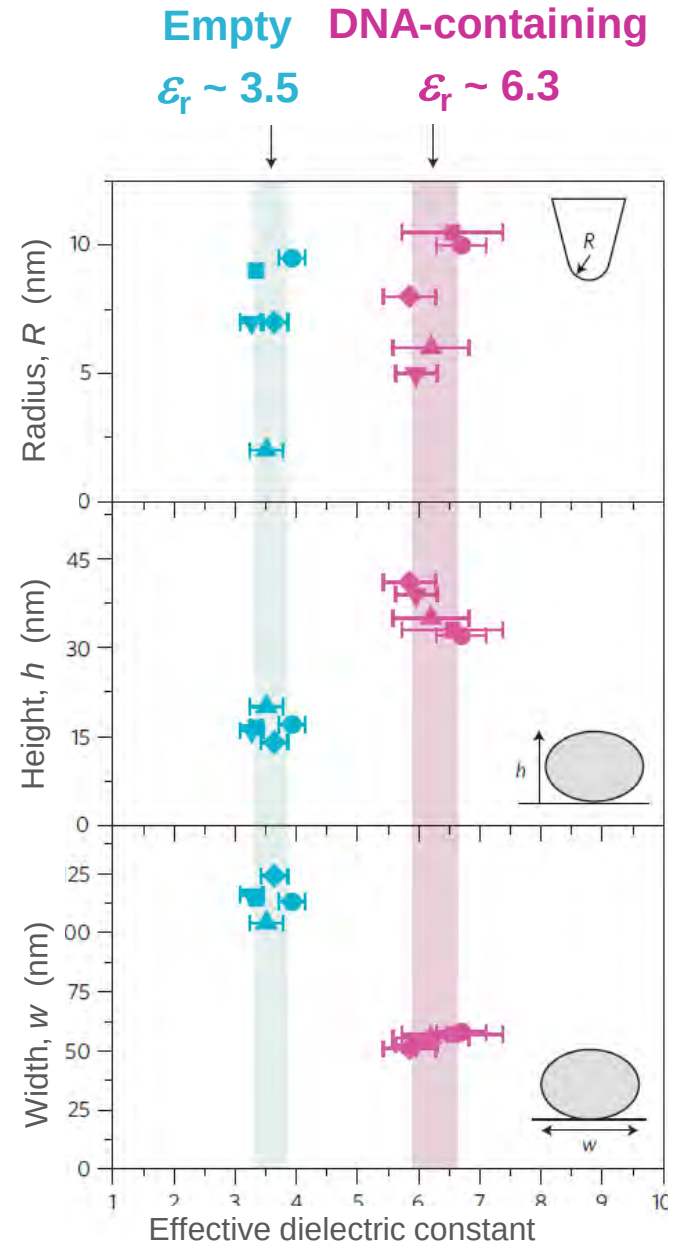
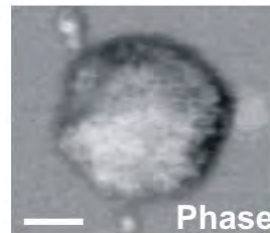
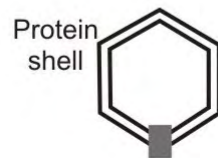
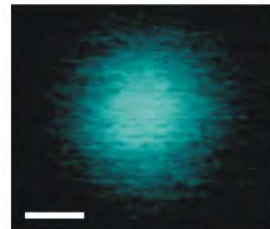
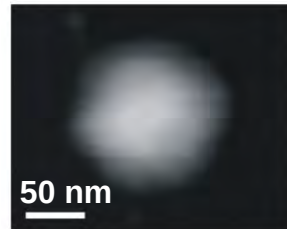
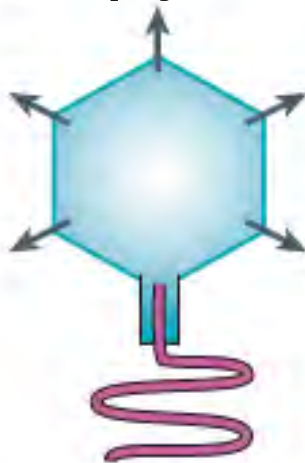
Independently from geometry
good precision $\delta\epsilon < 7\%$

A nanoparticle of unknown properties

Full virus



Empty virus

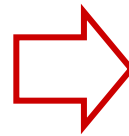


Dielectric polarization properties of DNA

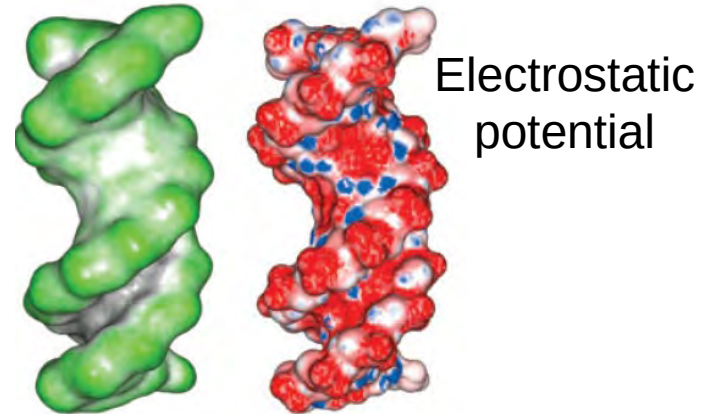
$$\nabla \cdot (\epsilon(\vec{r})) \nabla \phi(\vec{r}) = -[\rho_{\text{ions}}(\vec{r}) + \rho_{\text{perm}}(\vec{r})]$$

? Unknown

Complex structure
Interplay with solvent (water)

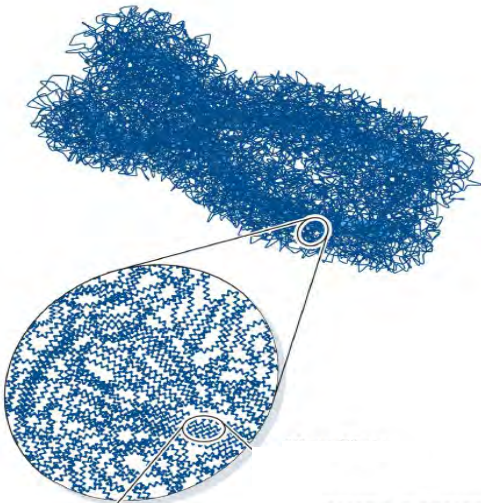


3D electrostatic potential maps depend on ϵ_{DNA} !



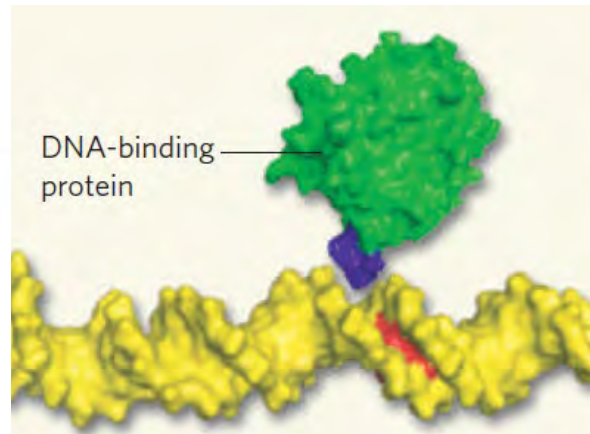
Major therapeutic impact

DNA structure

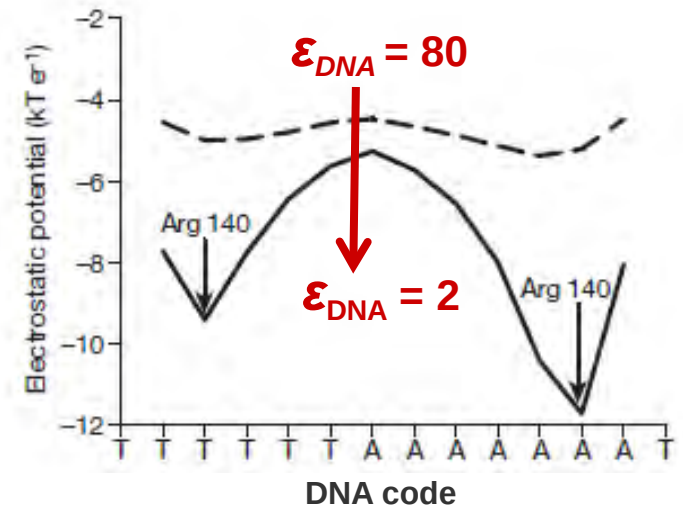


Molineux et al. Nature Rev. 2013

DNA-protein affinity



Rohs et al. Nature 2009

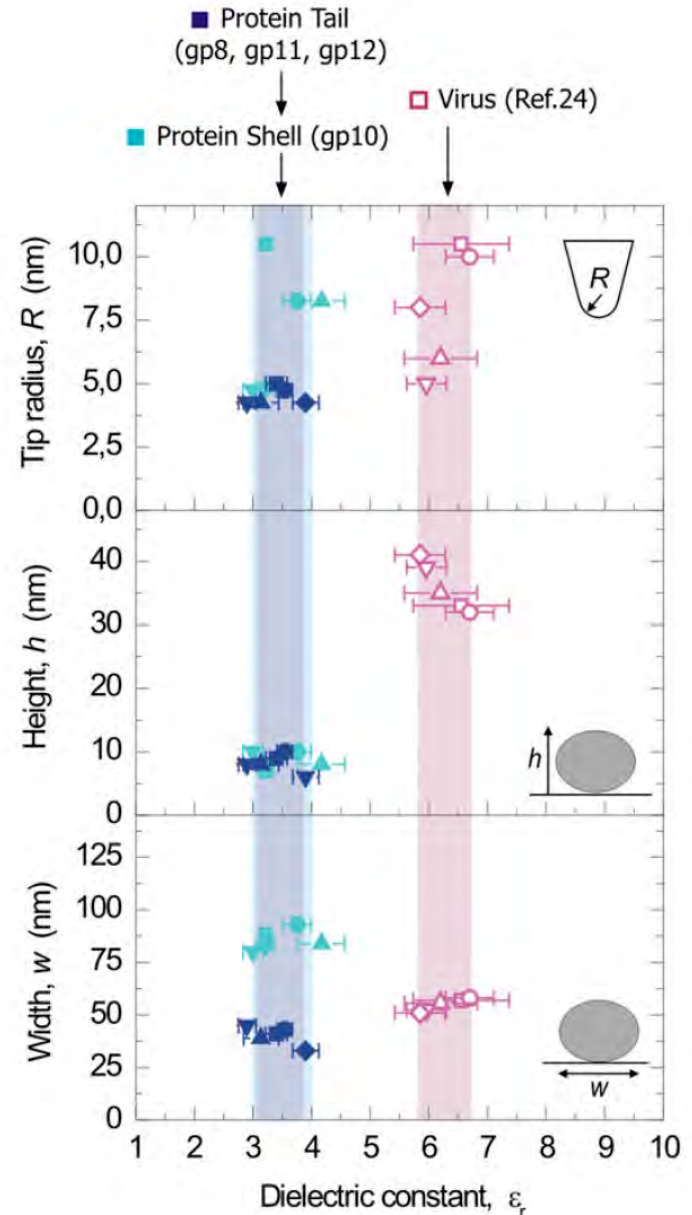
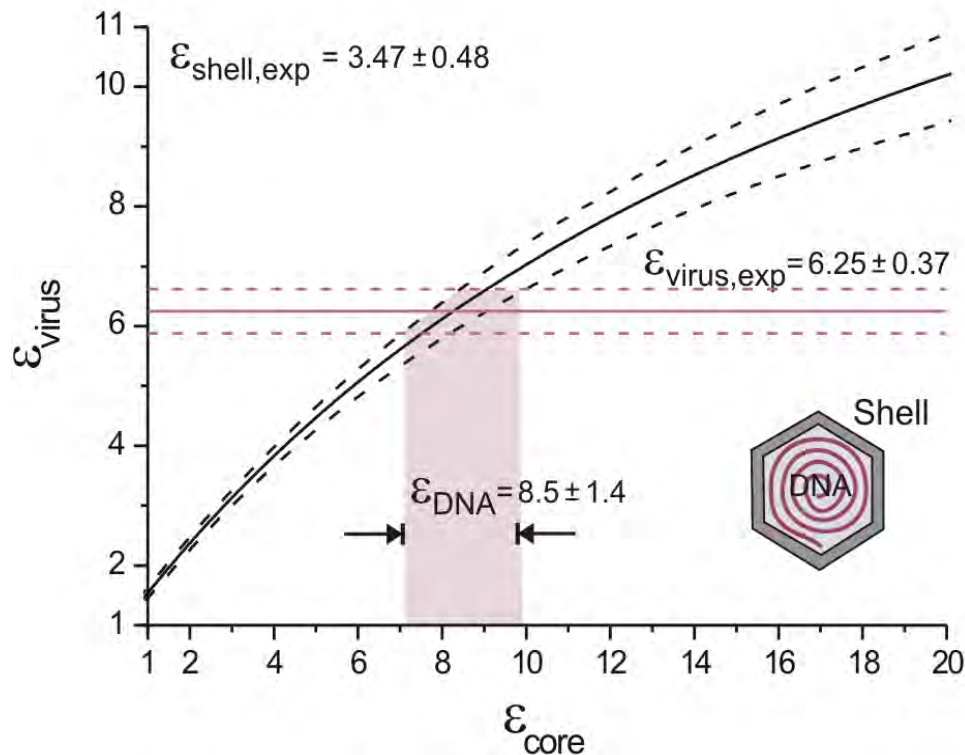


Dielectric polarization properties of DNA

Poisson-Boltzmann equation

$$\nabla \cdot (\epsilon(\vec{r}) \nabla \phi(\vec{r})) = -[\rho_{\text{ions}}(\vec{r}) + \rho_{\text{perm}}(\vec{r})]$$

$\epsilon_{\text{DNA}} = 8.5 \pm 1.4$ from our experiments
sensibly larger than normally assumed (2-4)

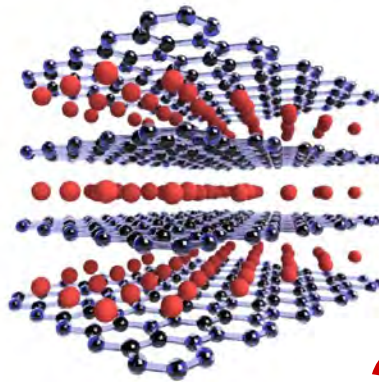
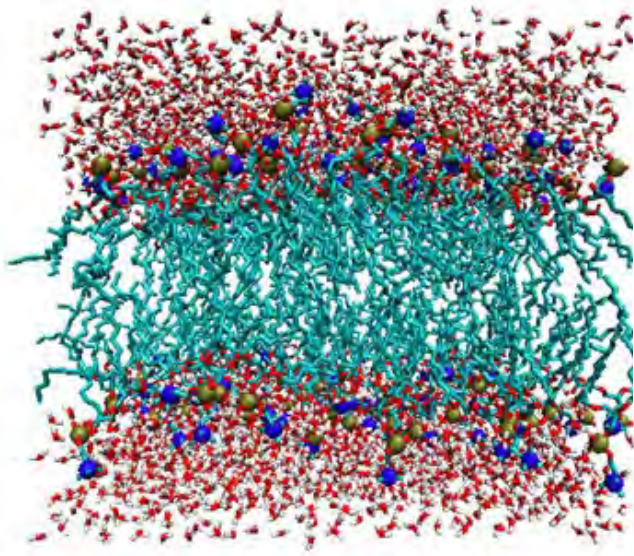


What is the dielectric constant of confined water?

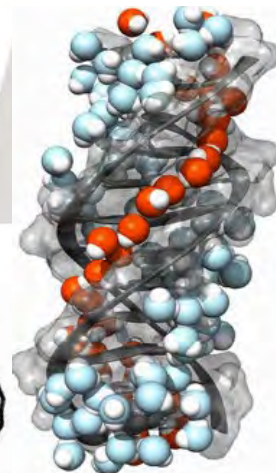
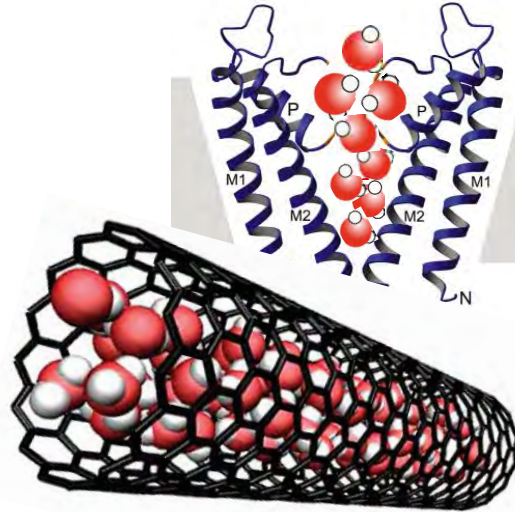
Remained unknown

No experimental tools with enough sensitivity

Interfaces



Nanocavities



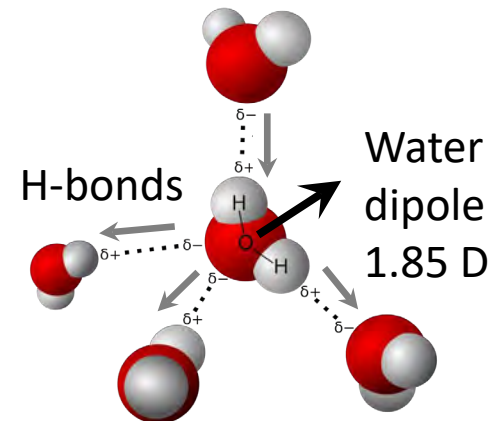
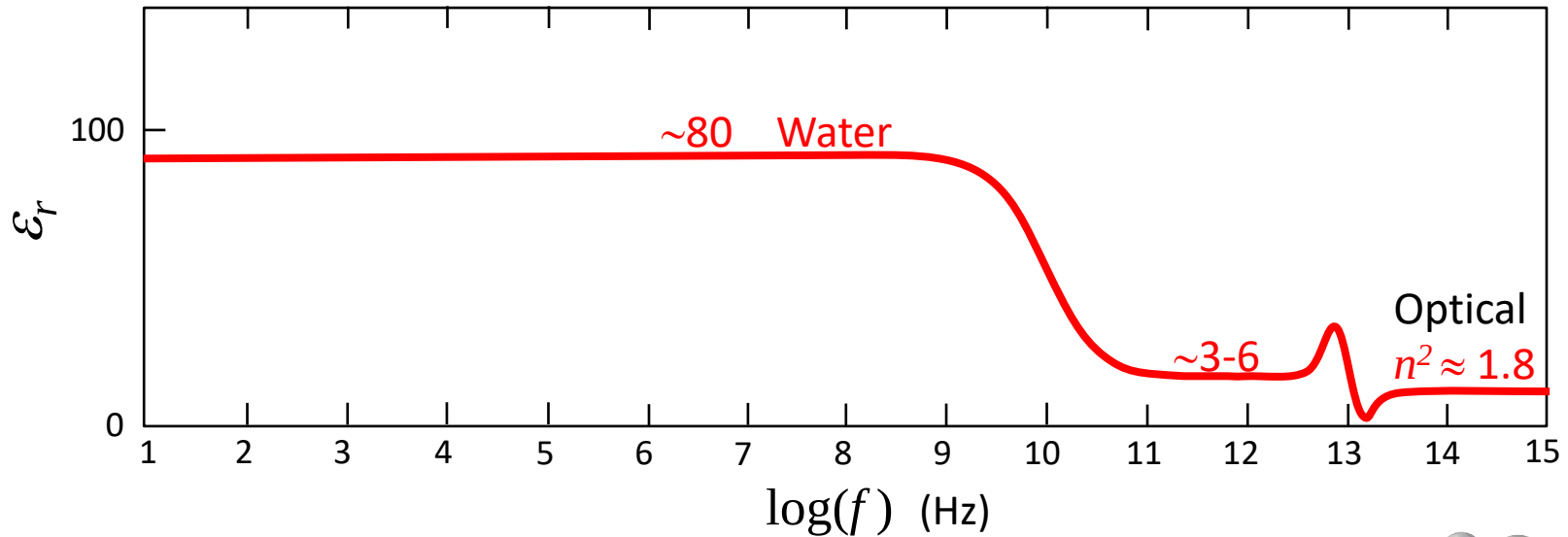
?
 ϵ_{water}

Ice-like hypothesis

Dielectric constant expected to be different from bulk

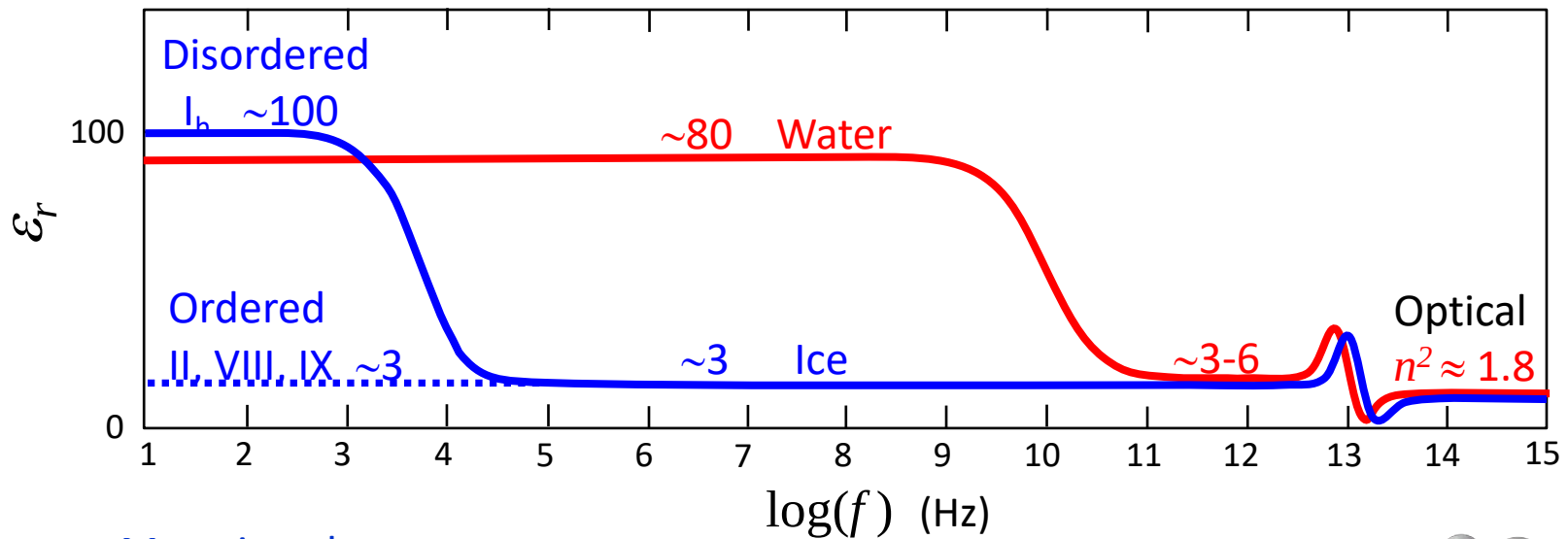
Dielectric spectra of bulk water and ice

Large ϵ_r due to water dipole and H-bonds polarization
→ ability to dissolve substances - “*the solvent of Life*”



Dielectric spectra of bulk water and ice

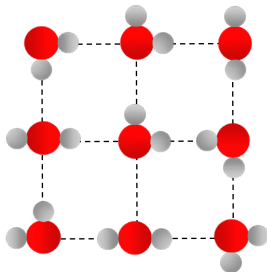
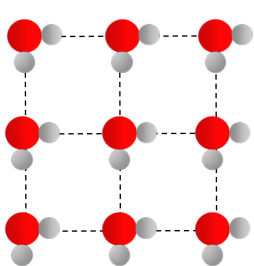
Large ϵ_r due to water dipole and H-bonds polarization
→ ability to dissolve substances - “*the solvent of Life*”



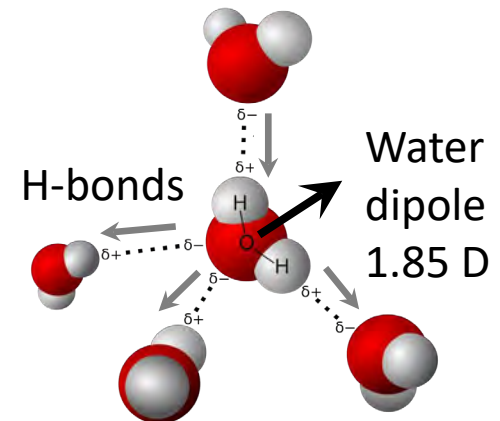
Many ice phases

H-ordered
small ϵ_r

H-disordered
large ϵ_r



Critical role played
by hydrogen ordering



One-century long experimental challenge

Complex geometries, too many uncertainties at large scale

Interfacial layer thickness ?

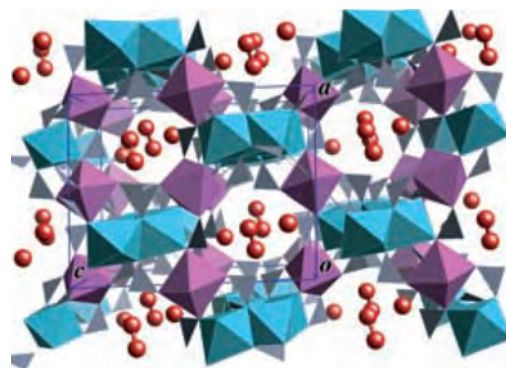
Electric double layer

Stern 1924
Conway 1951
Bockris 1962
Onsager 1977
etc.

primary water
layer, $\epsilon = 6$

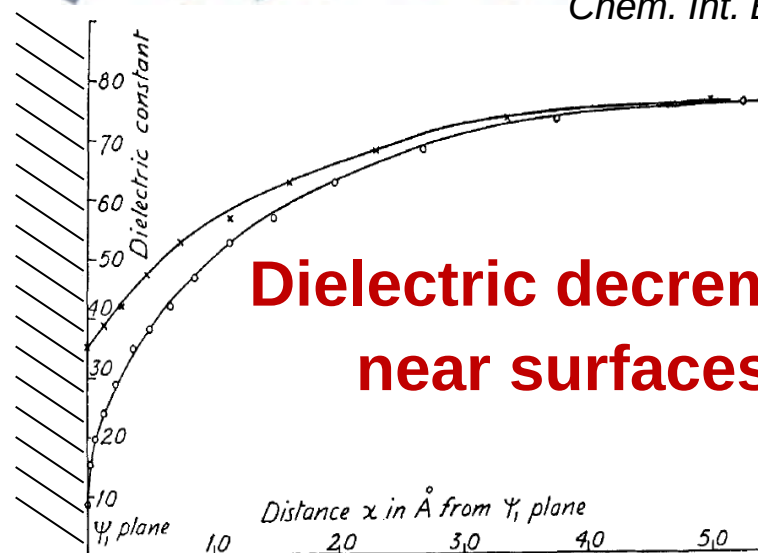
secondary water
layer, $\epsilon = 32$

$\epsilon = \infty$



Nanoporous
crystals,
powders,
dispersions

Cui et al. *Angew.
Chem. Int. Ed.* 2005



**Dielectric decrement
near surfaces**

FIG. 3.—Dielectric constant against distance curves for the diffuse double layer for concentration = 0.01 g. ion/l.

× $\psi = 0.1$ V ○ $\psi = 0.2$ V
× $\psi = 0.1$ V ○ $\psi = 0.2$ V

Bockris et al. *Proc. R. Soc. Lond. A* 1963

New approach: use of 2D atomically thin crystals

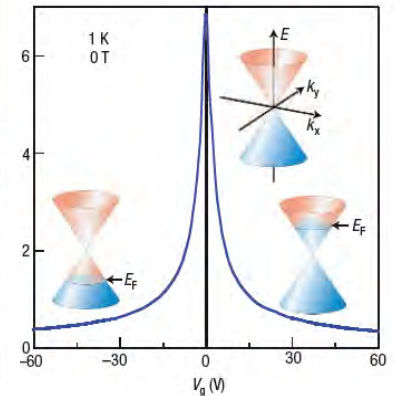
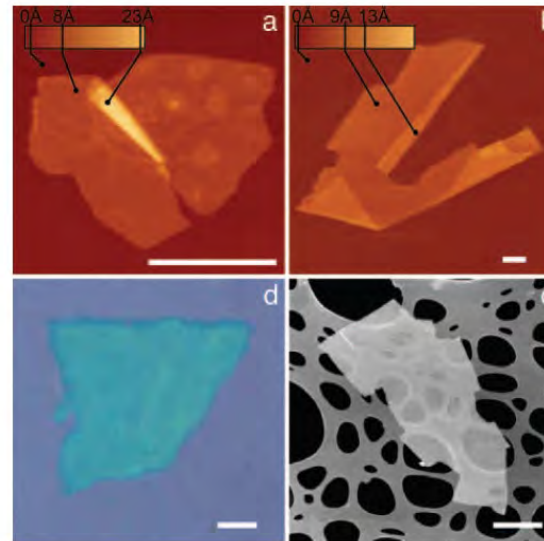


School of Physics & Astronomy
National Graphene Institute

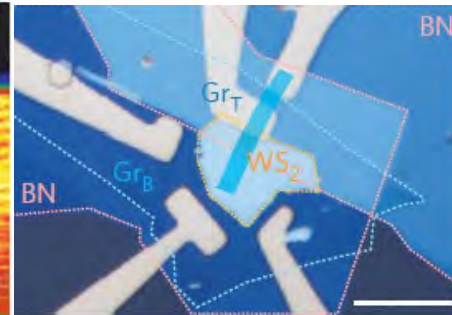
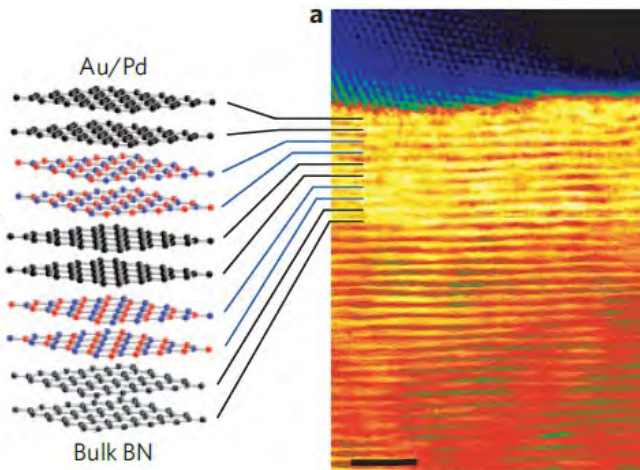


Graphene and
other van der Waals crystals

Novoselov & Geim
Science, *Nature* 2004
PNAS 2005



van der Waals
heterostructures

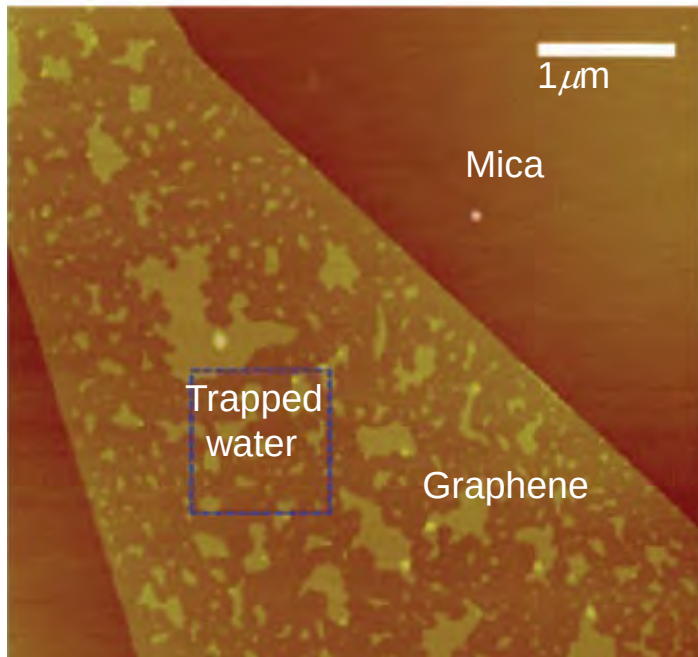


Geim and Novoselov,
Nobel Prize in Physics, 2010

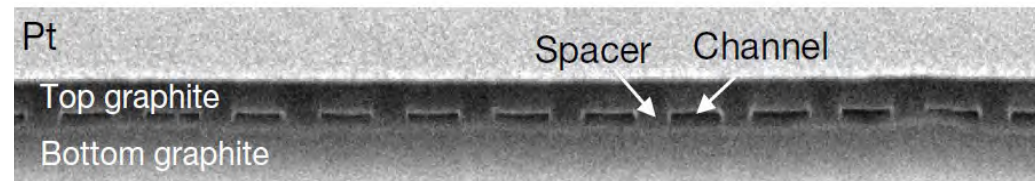
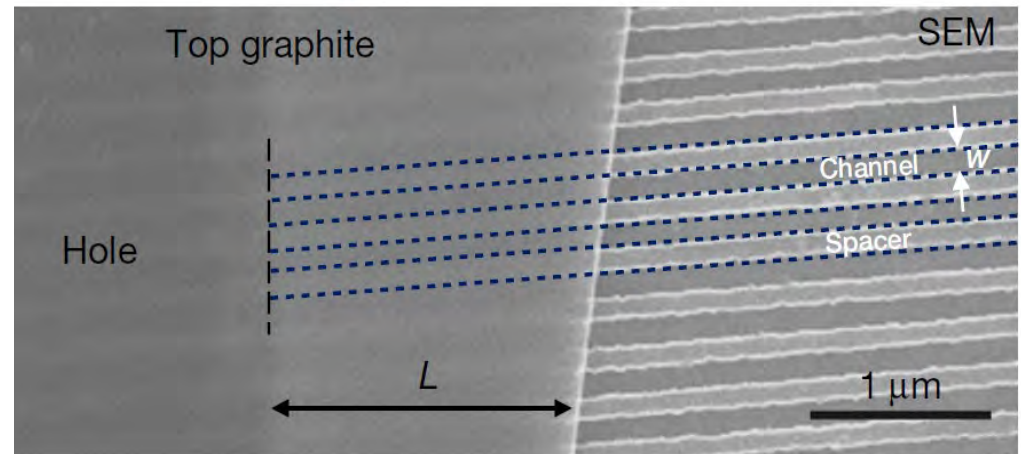
Water confined by van der Waals crystals

Ideal experimental platform

Atomically flat and smooth interface



Xu et al. Science 2010



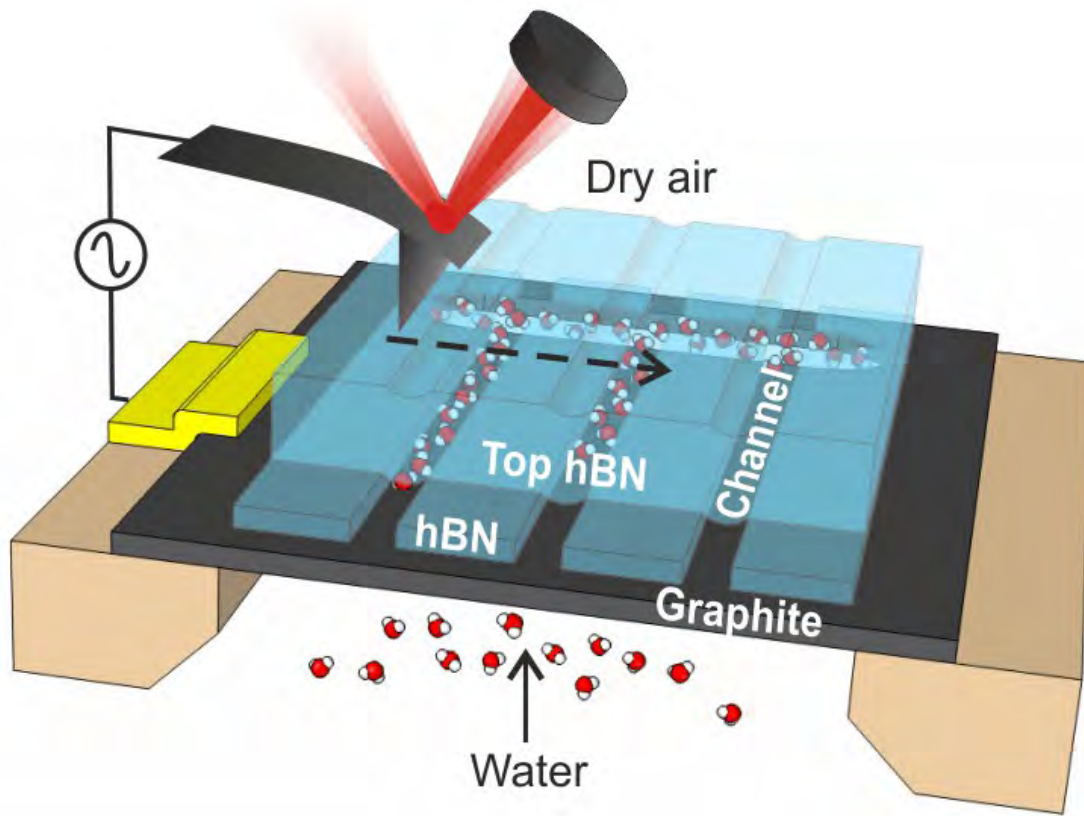
Radha et al. Nature 2016
Esfandiar et al. Science 2017

**A. K. Geim's group
(Manchester)**

Dielectric imaging of single nanochannels

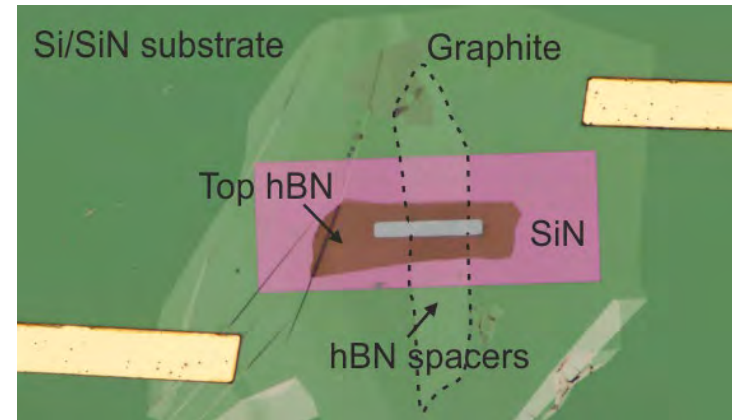
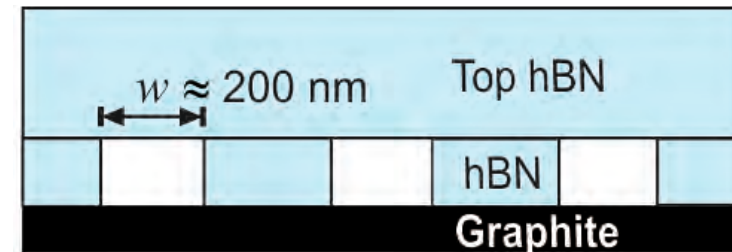
hBN/Graphite channels

hBN as electrically transparent top layer



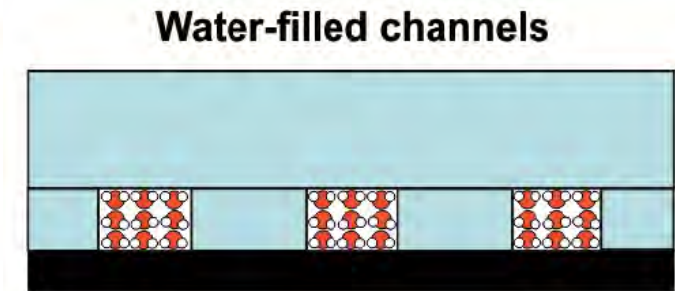
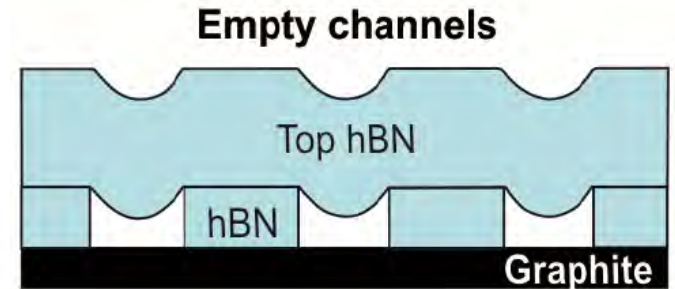
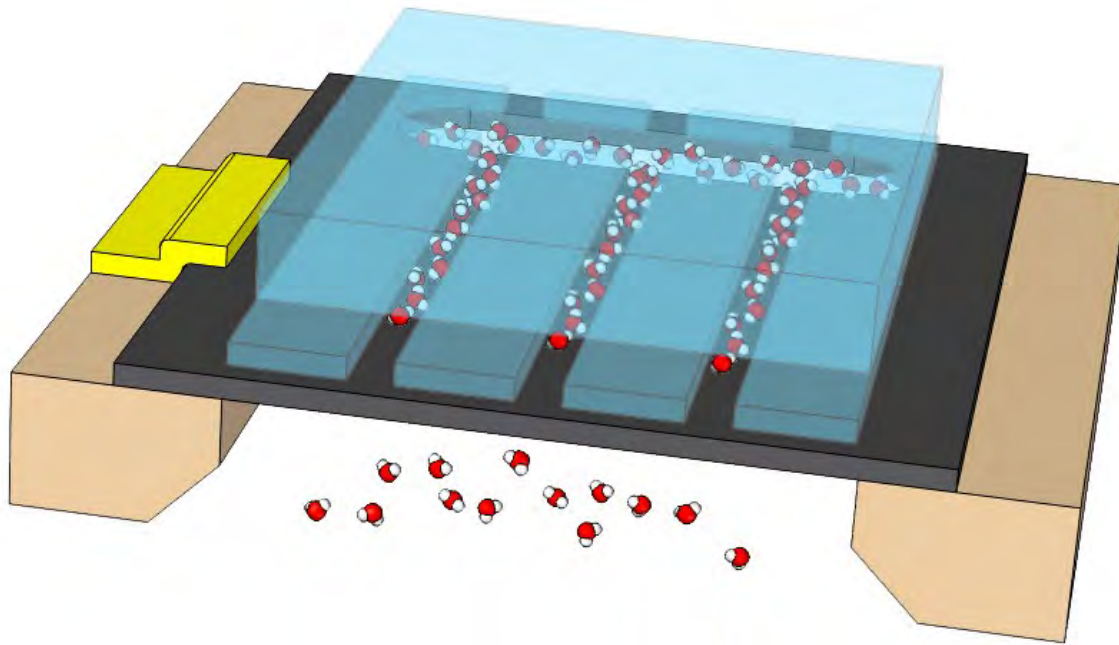
NB hBN spacers

*Reference for $\epsilon_r \sim 3.5$
(ordered ice)*



Device built with sagged top layer

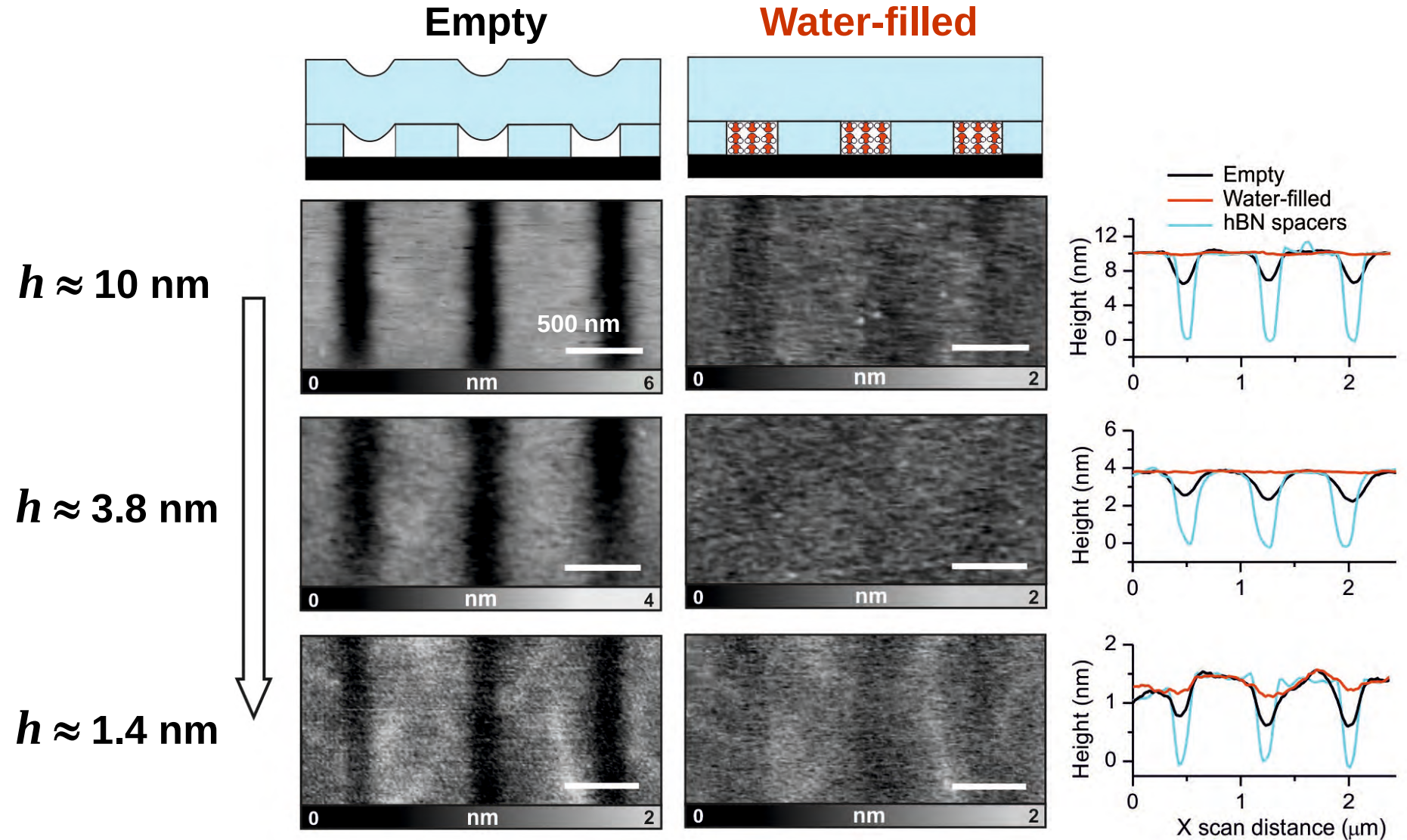
Fine-tuning of the top-layer thickness to avoid total collapse



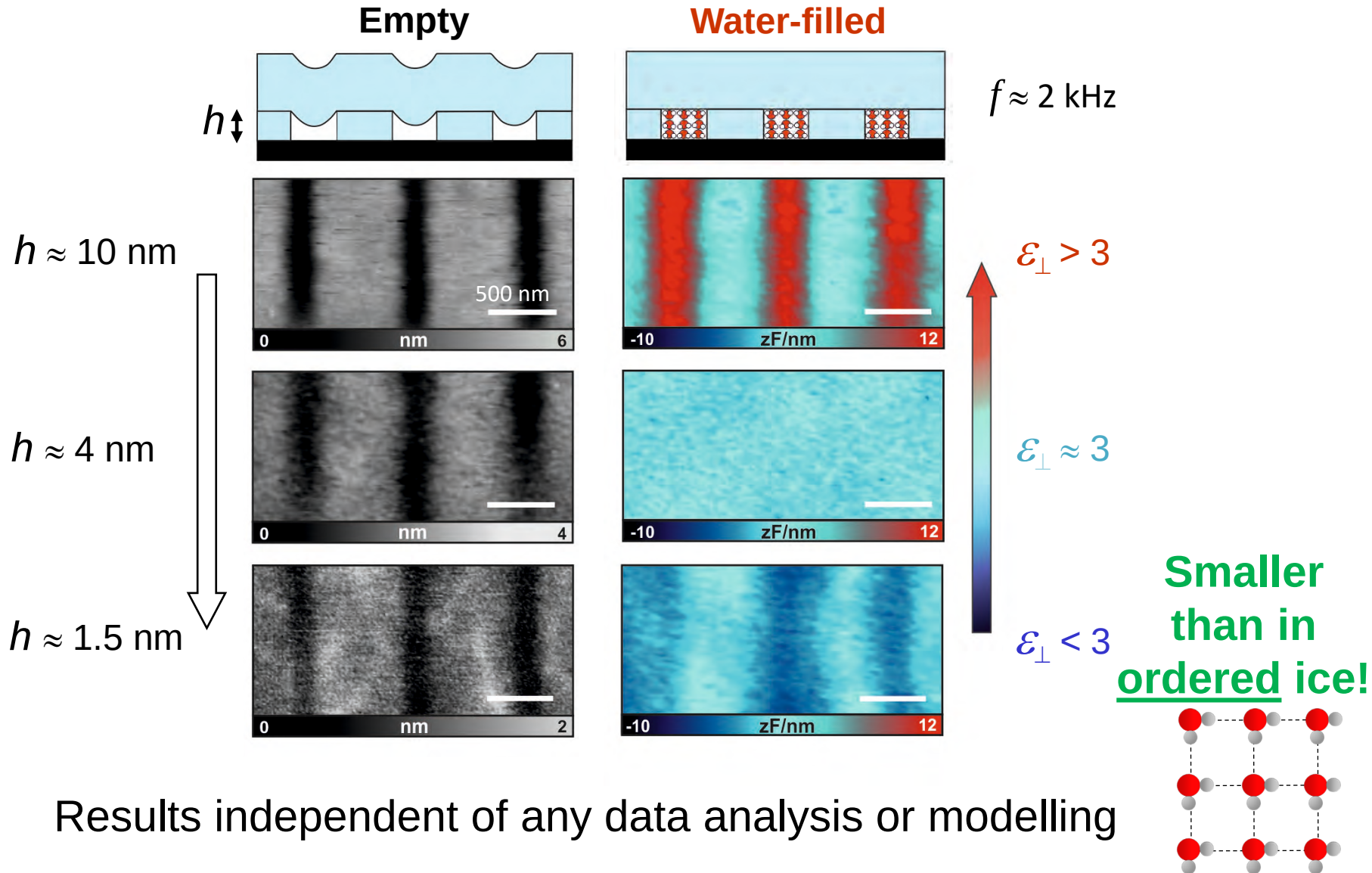
The top layer becomes straight when the channel is filled

Water

Nanochannels' thickness down to ~ 1 nm



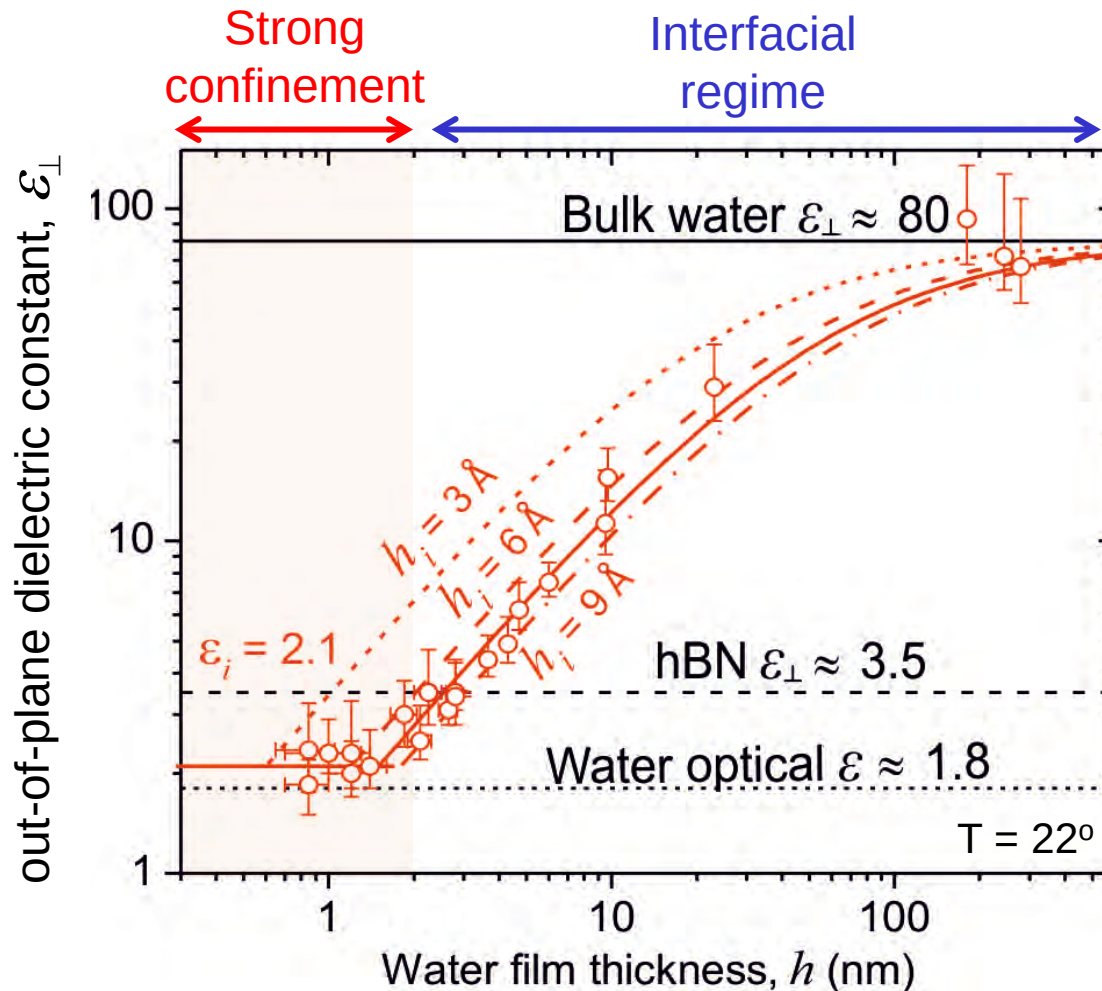
Anomalous low out-of-plane ϵ_r of 2D water



Results independent of any data analysis or modelling

Out-of-plane ϵ_r of 2D confined water

Non-polarizable: dipolar and H-bond polarization are **suppressed**



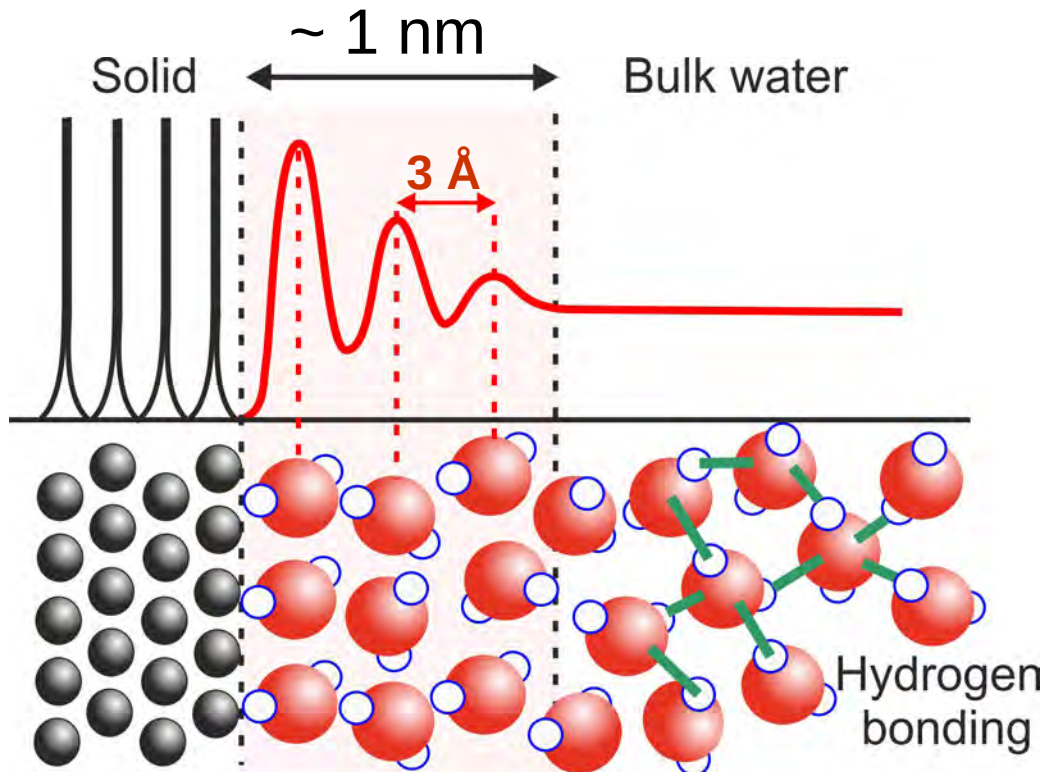
TWO REGIMES
with same behavior

Residual
electronic and atomic
polarization

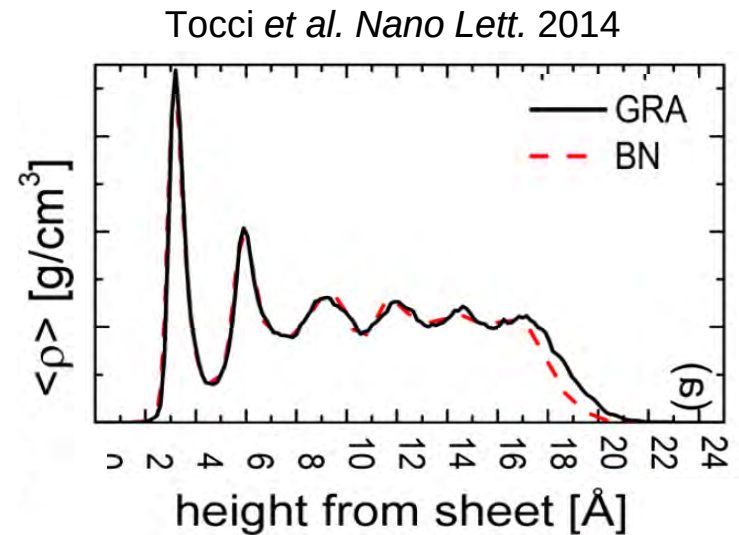
← $\epsilon_{\perp} \approx 2$
near the optical limit

Layered structuring of water near surfaces

Molecular Dynamics density calculations



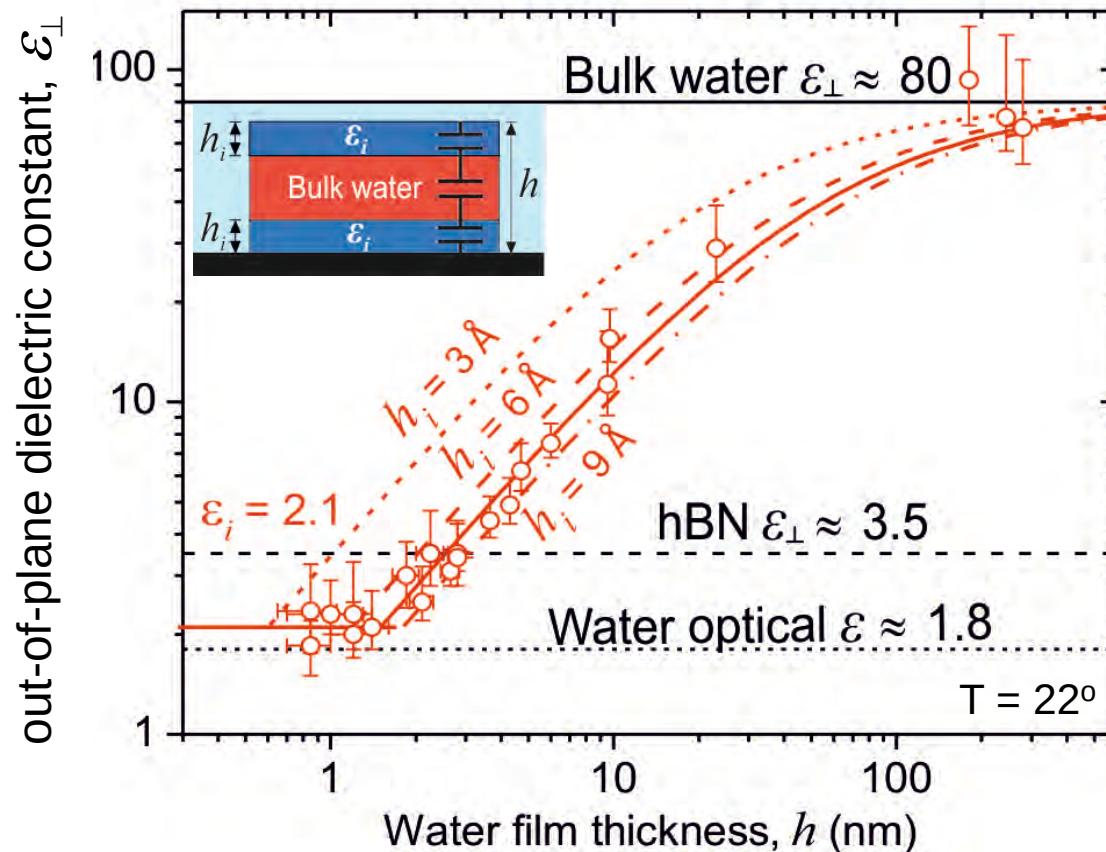
***2-3 ordered layers
are predicted***



**Same behaviour
for hBN and graphite**

Out-of-plane ϵ_r of 2D confined water

Electrically dead layer with $\epsilon_{\perp} \approx 2$ extending ~ 2 water layers
Same behavior for interfacial and strongly confined water



Consistent with most
MD predictions

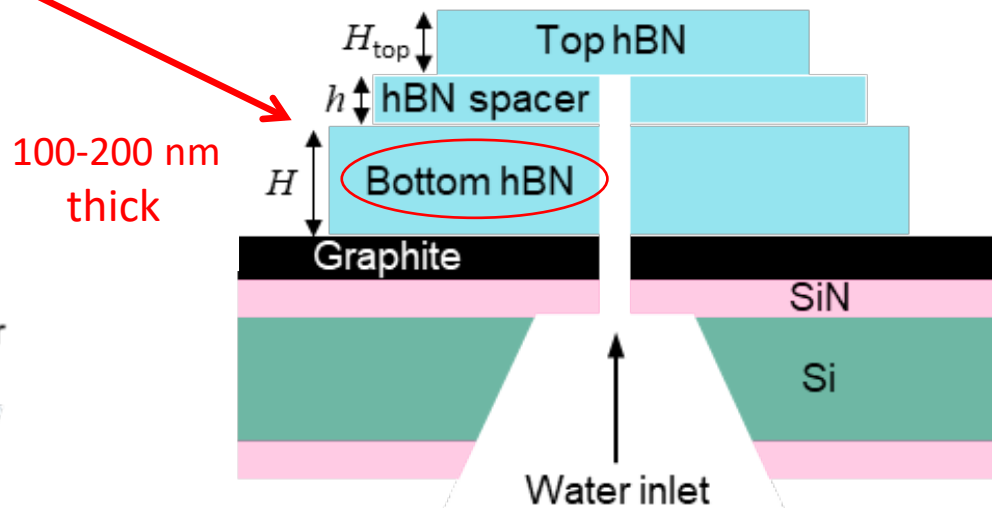
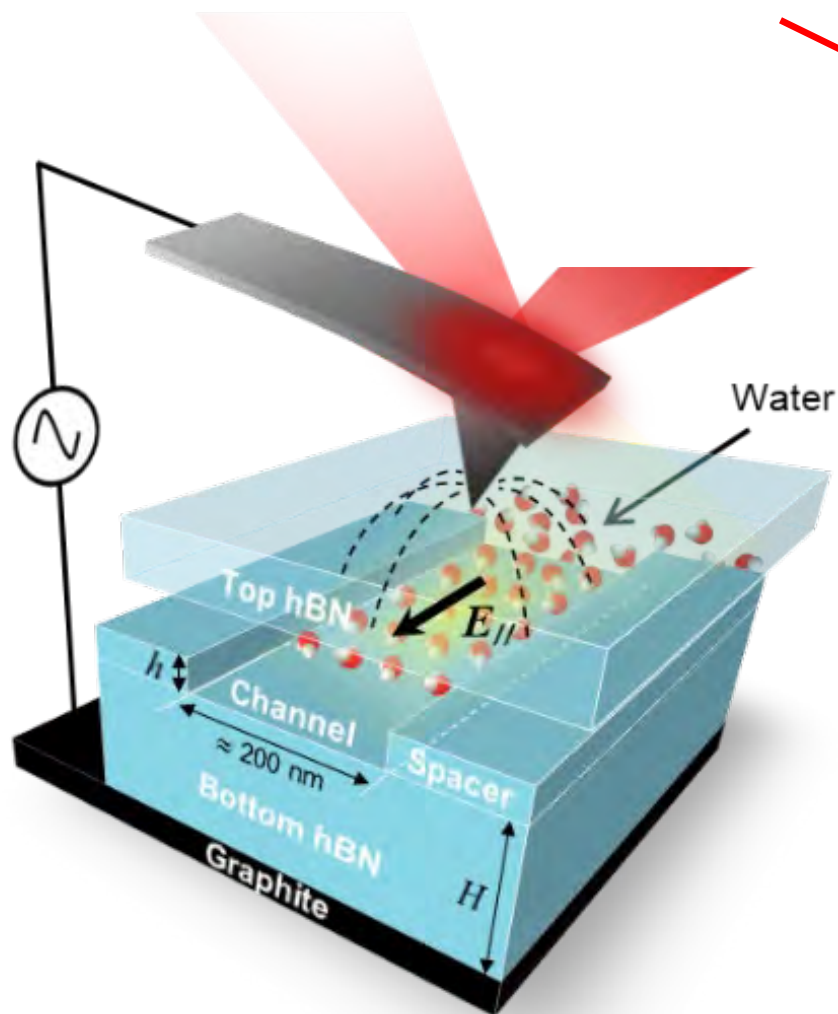
Netz 2011
Galli 2013
Aluru 2020
Marx 2020
Peeters 2021
Matyushov 2021
and many others

Question

**Can we probe the in-plane dielectric constant
of 2D confined water?**

In-plane dielectric imaging of 2D confined water

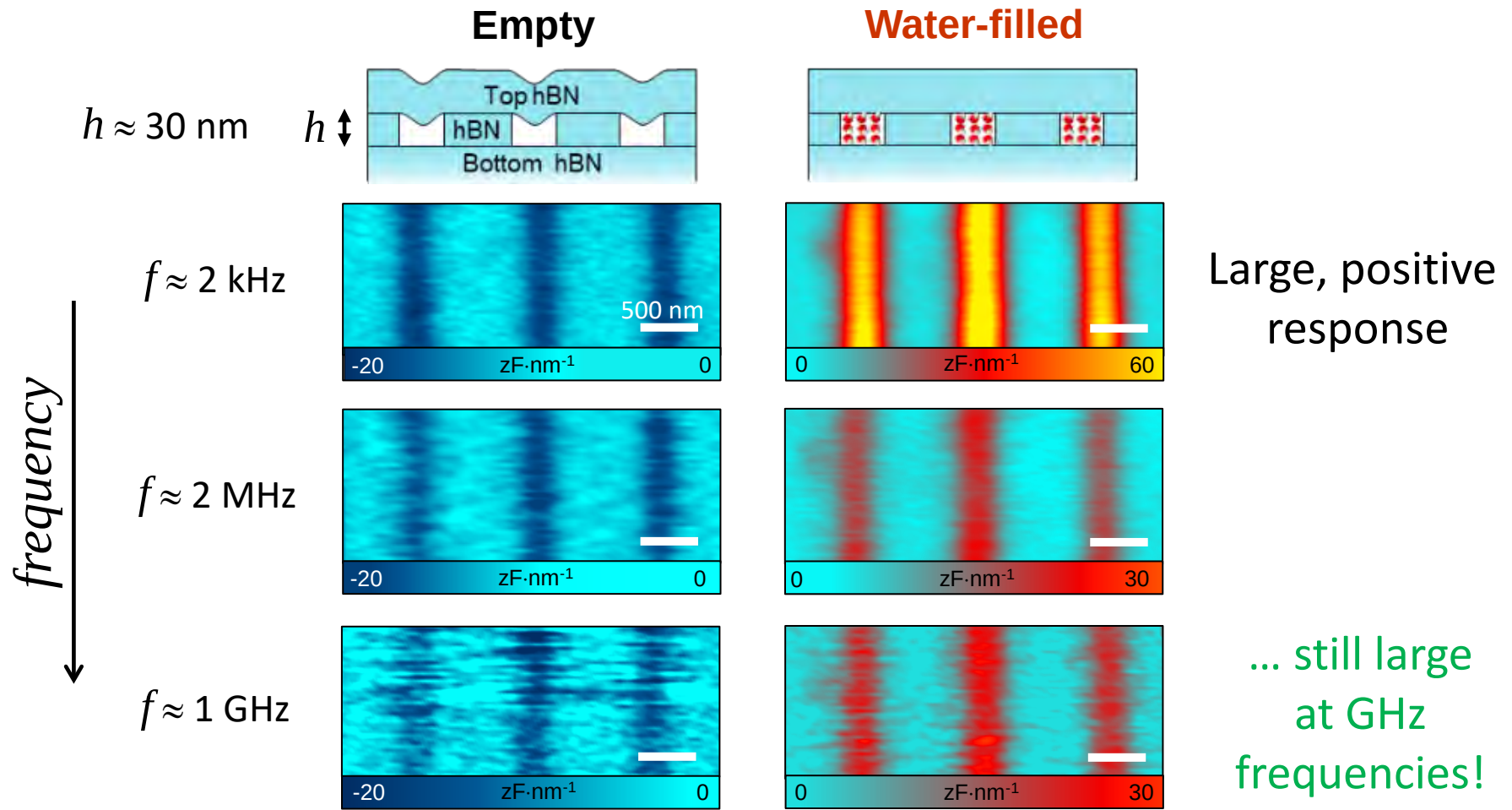
In-plane electric field inside the channel



M. Souilamas, R. Wang



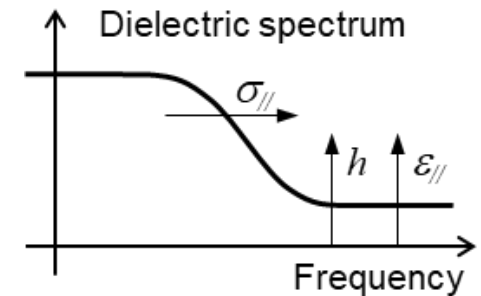
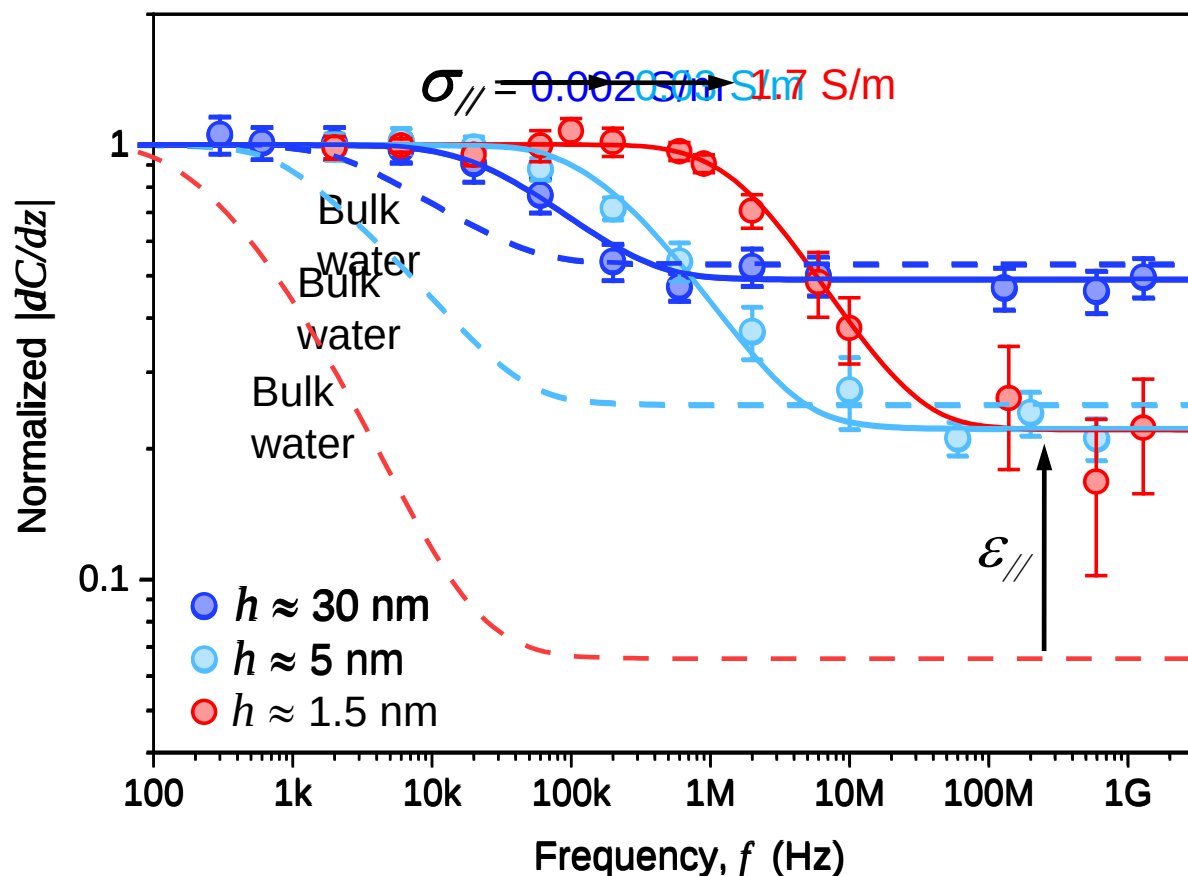
In-plane dielectric imaging of 2D confined water



Decreasing the thickness down to atomic scale

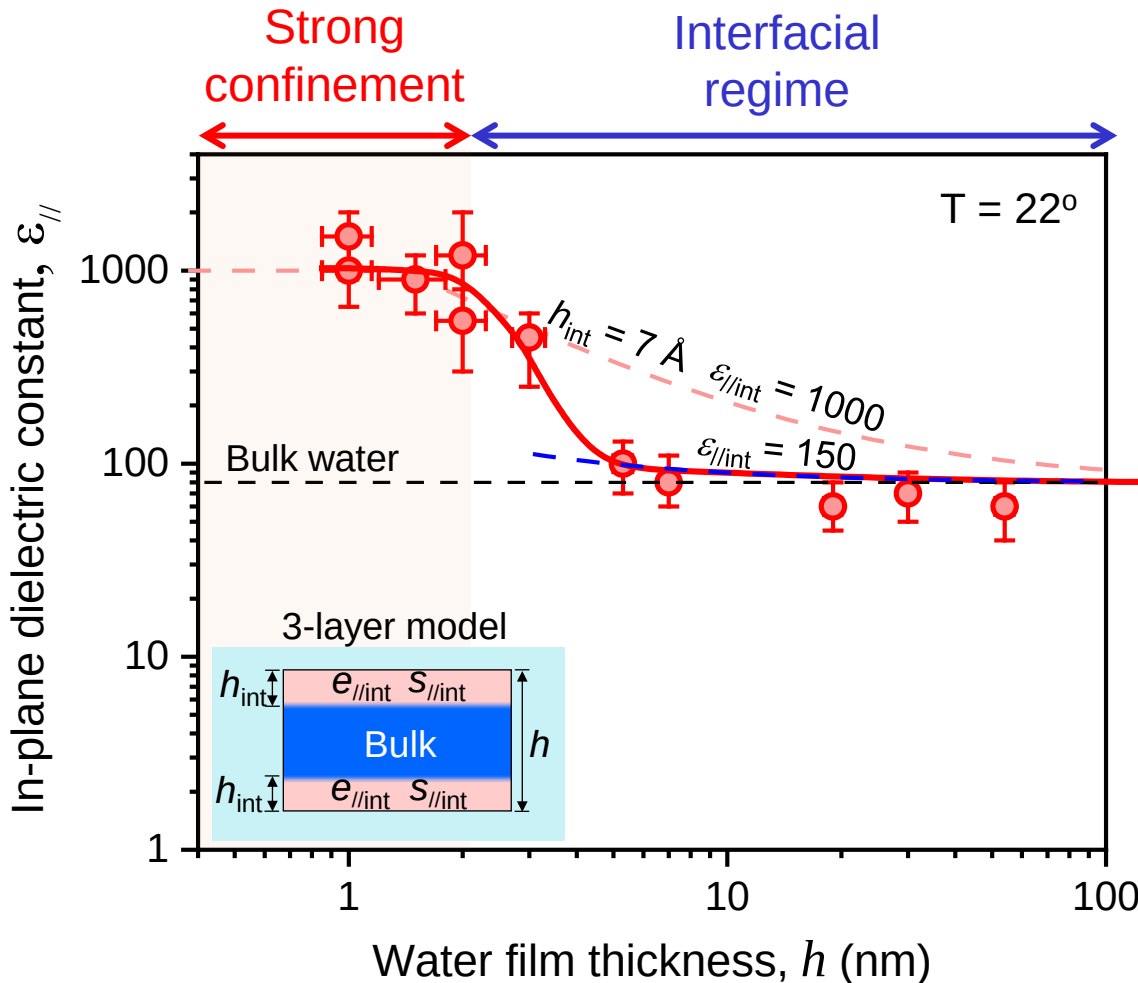
The dispersion clearly shifts → **Anomalously high $\sigma_{//}$ and $\epsilon_{//}$**

Results independent of any data analysis or modelling



In-plane ϵ_r of 2D confined water

Highly polarizable: values increase with confinement strength



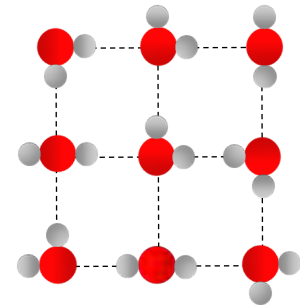
TWO REGIMES

with different behavior

Bulk-like or slightly higher
in the interfacial regime

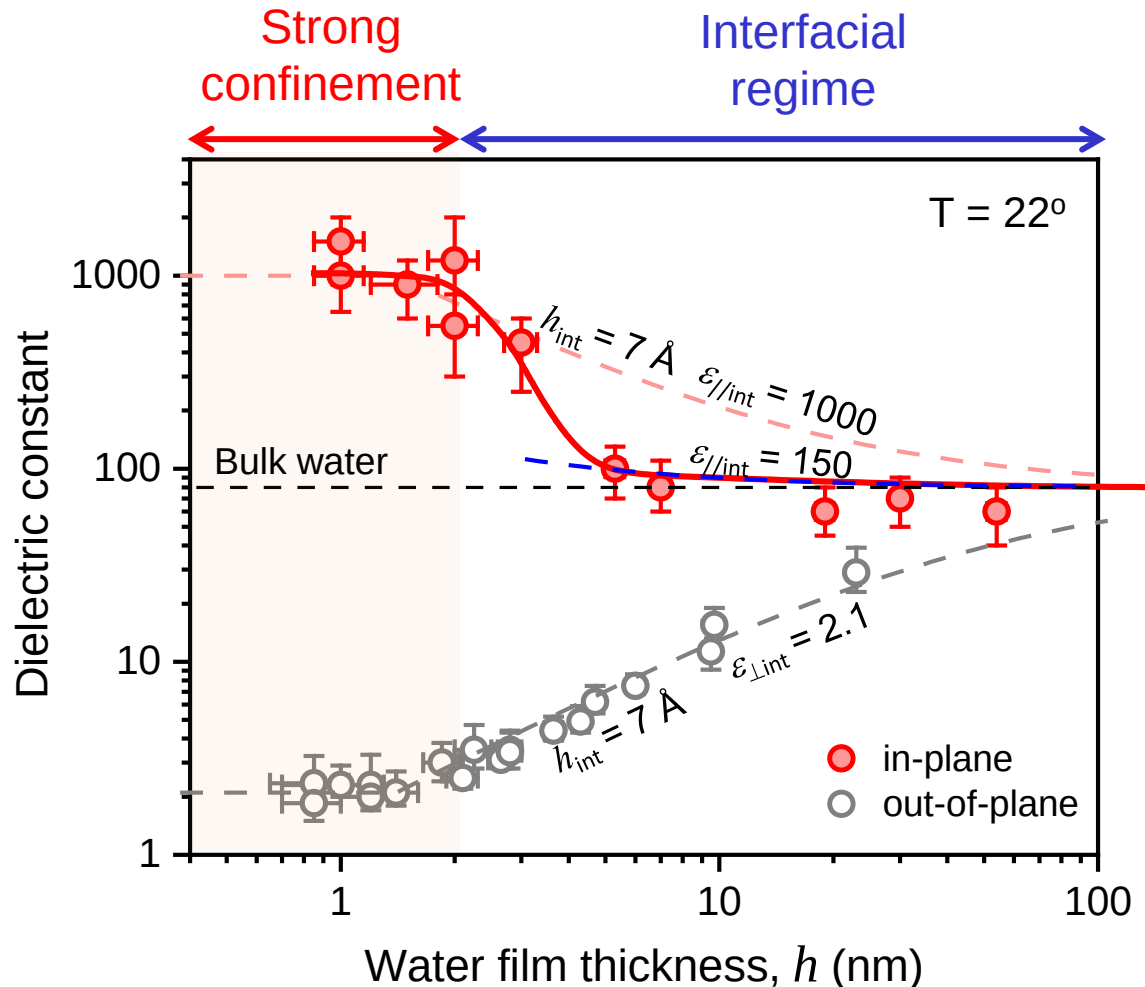
Ferroelectric-like values
under strong confinement

→ Higher than in
disordered ice



Dielectric constant of 2D confined water

Giant anisotropy in dielectric screening



Consistent with most
MD predictions
(at least qualitatively)

[Netz 2011](#)

[Galli 2013](#)

[Aluru 2020](#)

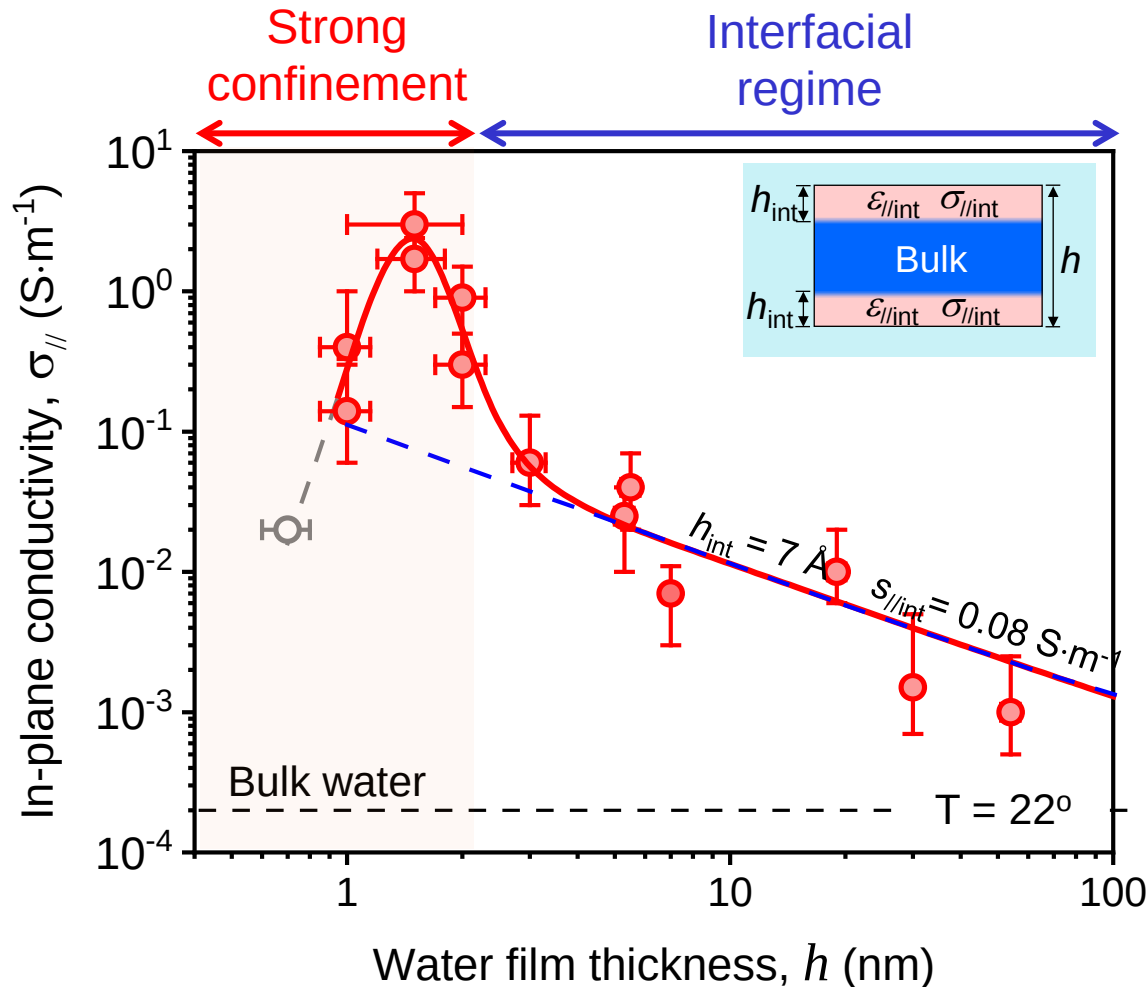
[Marx 2020](#)

[Peeters 2021](#)

[Matyushov 2021](#)
and many others

In-plane conductivity of 2D confined water

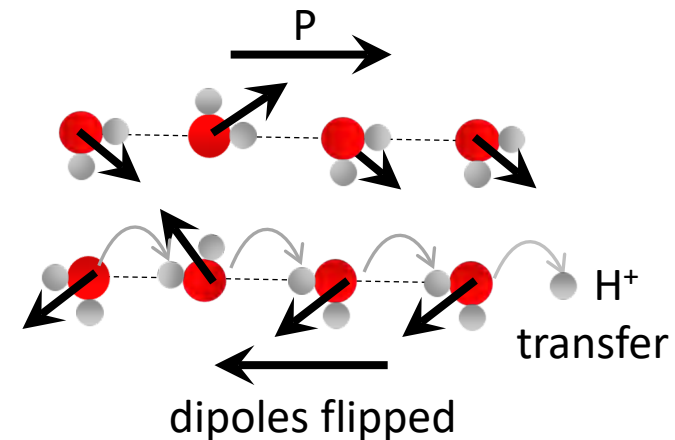
Increase up to **superionic values** under strong confinement



TWO REGIMES
with different behavior

Under strong confinement,
similar to disordered ice:

H-disorder facilitates
dipole-dipole correlation
and Grotthuss-type
fast proton transport



Question

**Can we probe local electric polarization
in liquid environment?**

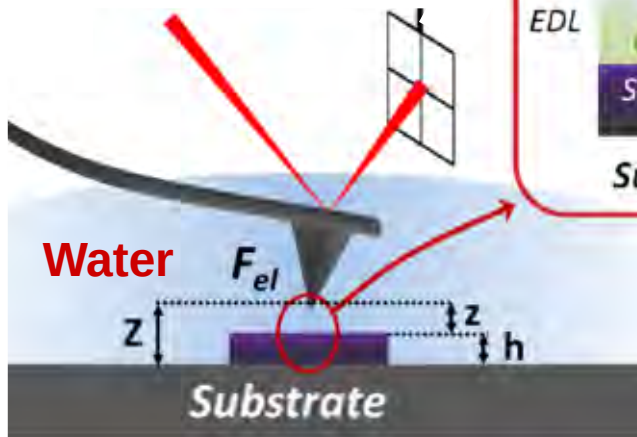
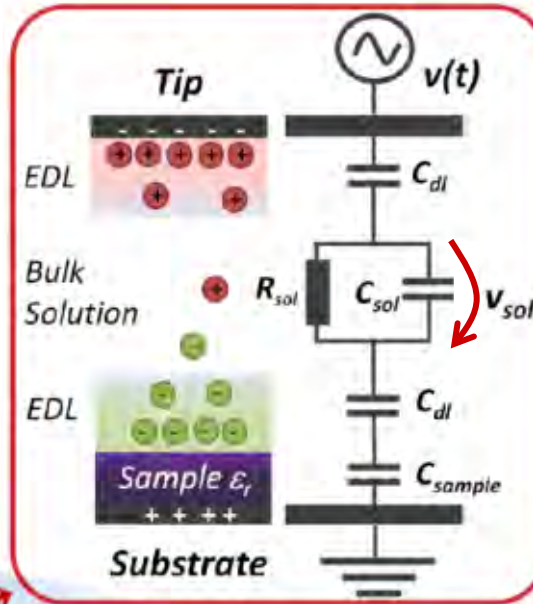
EFM in electrolyte solutions

$$F_{elec}(z, t) = \frac{1}{2} \frac{\partial C_{sol}(z)}{\partial z} v_{sol}^2(z, t)$$

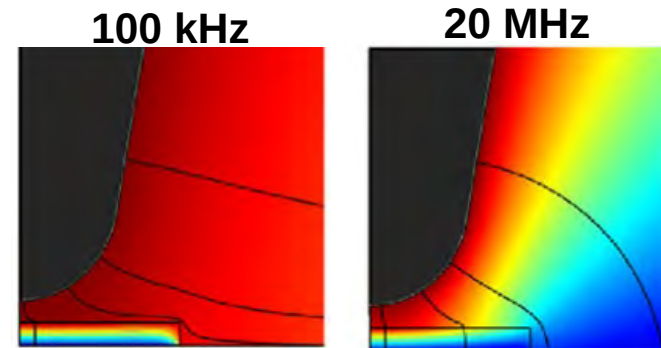
The voltage drop depends on frequency and ion concentration

Advantage: higher dielectric environment $\epsilon_r \sim 80$

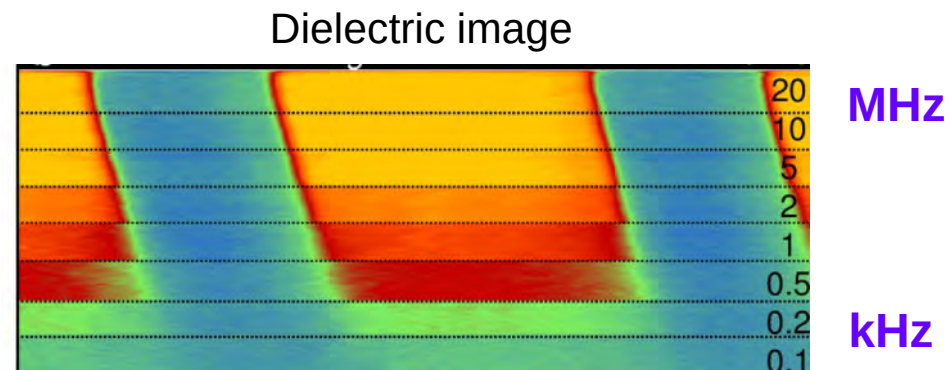
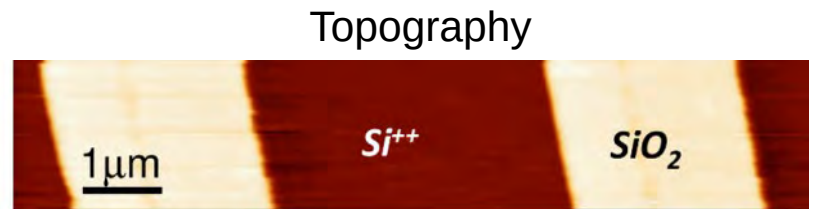
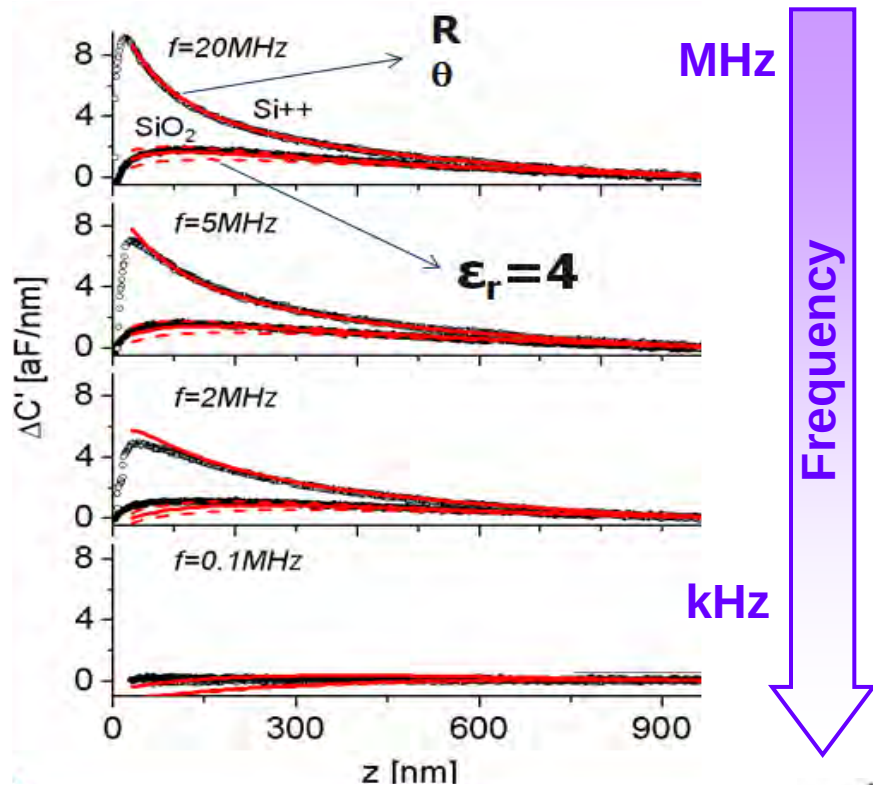
Issues: ions conductivity and electric double layer



Tip apex interaction only at frequency > 1MHz

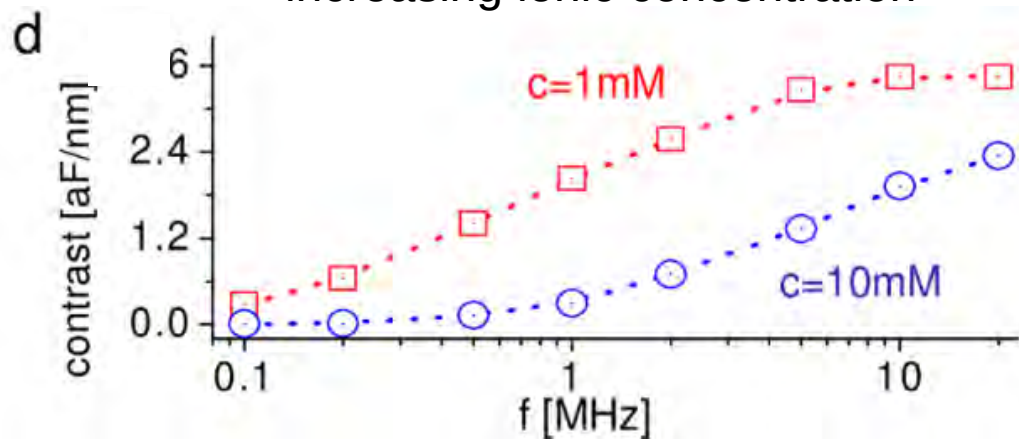


EFM in electrolyte solutions at MHz frequencies



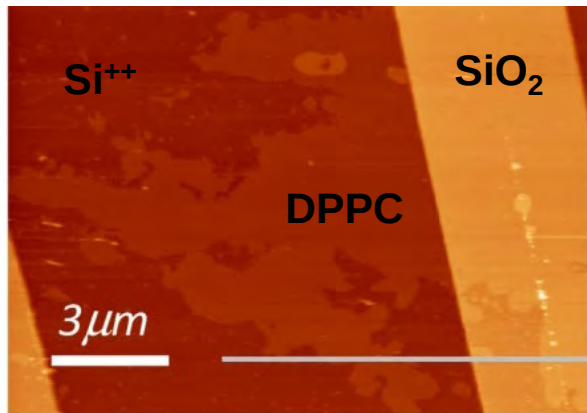
**At high frequencies > MHz
it works as in air !**

Higher frequencies with
increasing ionic concentration

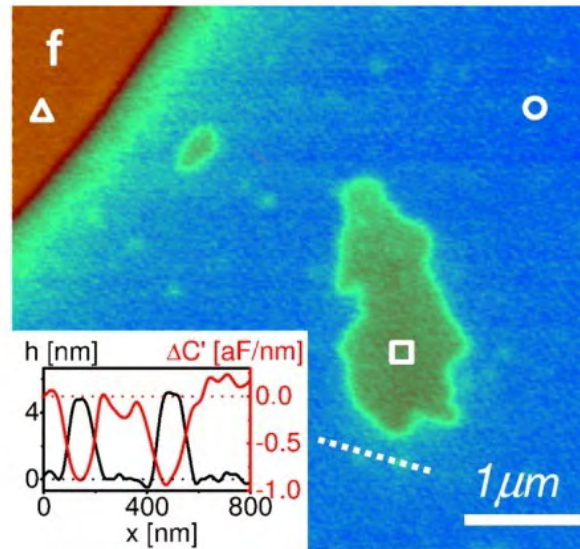
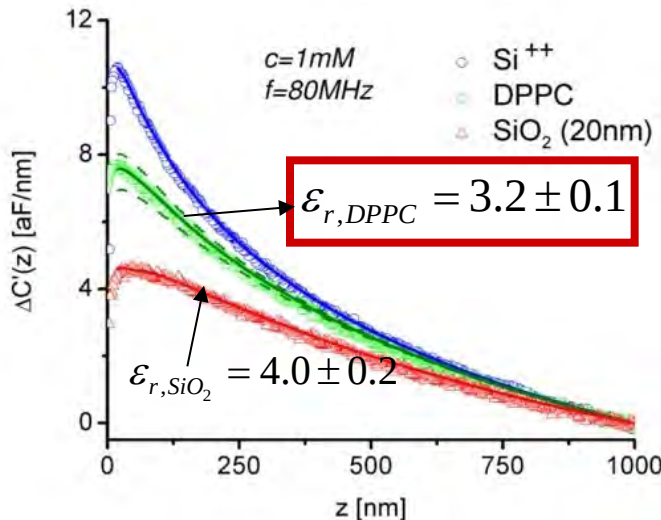
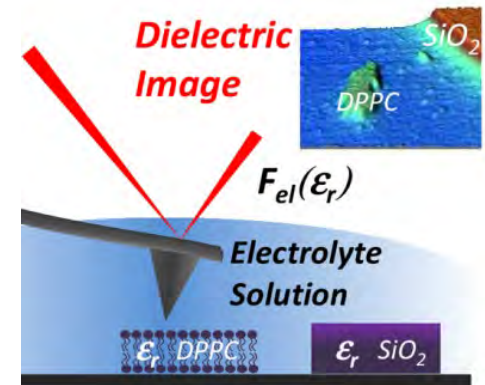
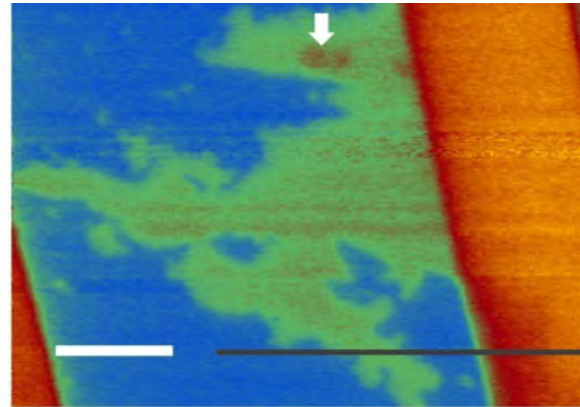


Dielectric constant of lipid bilayers in liquid

Topography



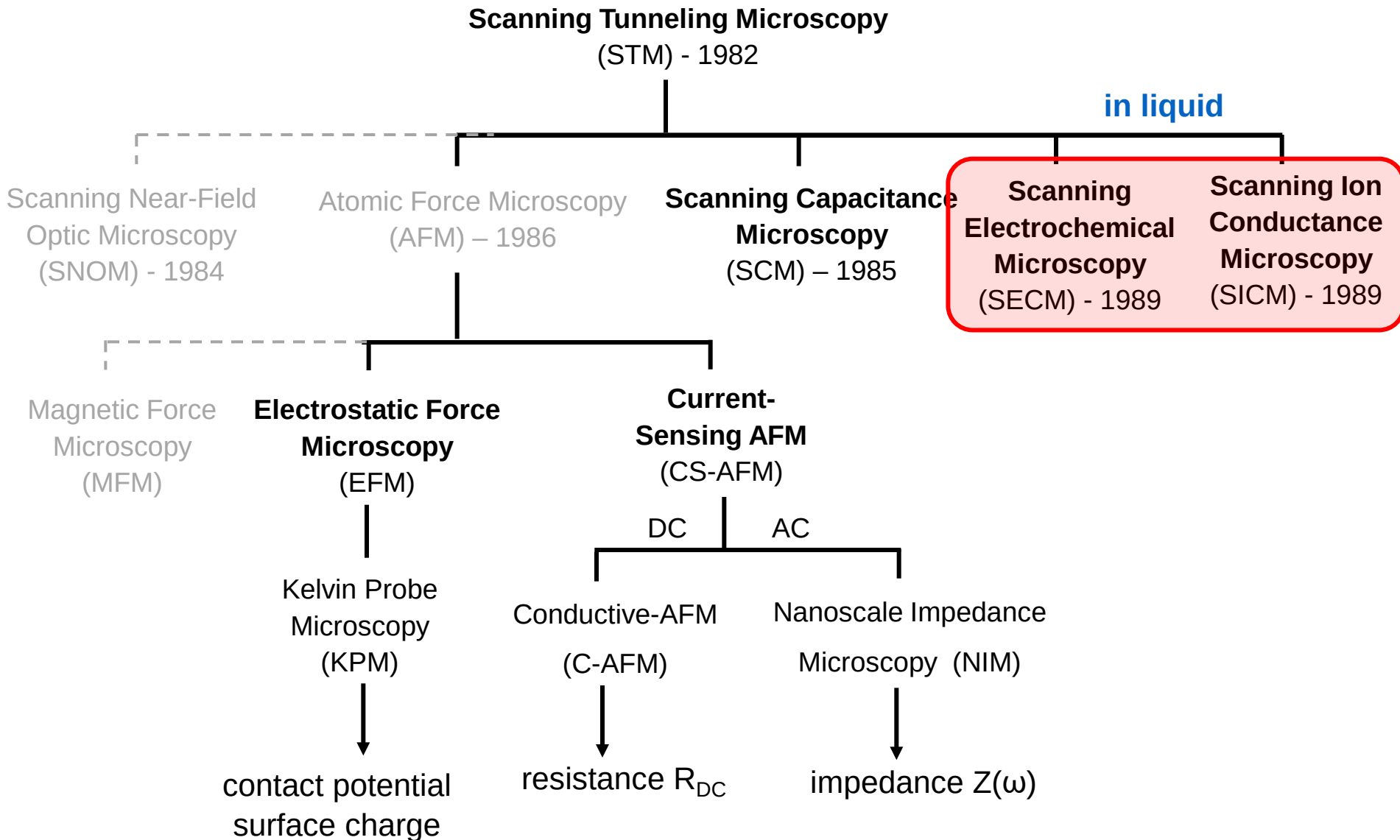
Dielectric image



$$\rightarrow \epsilon_r \sim 3$$

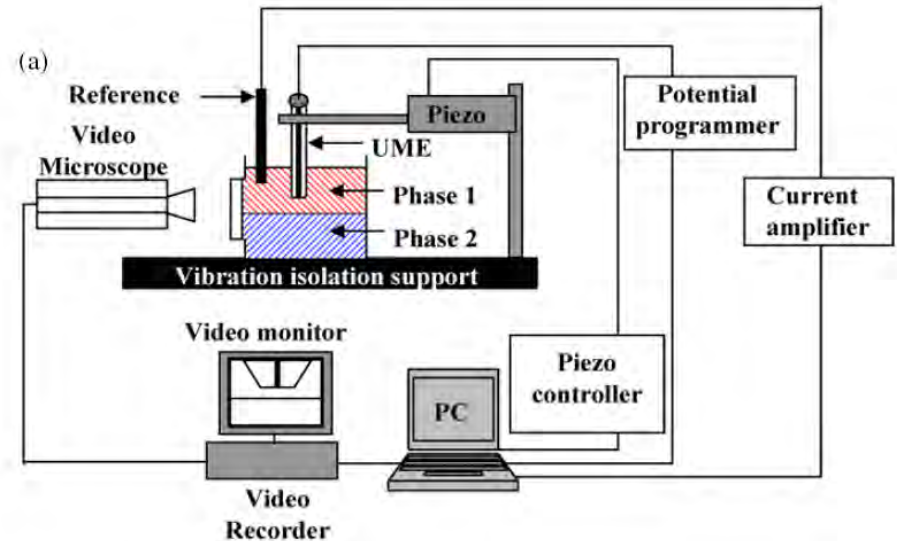
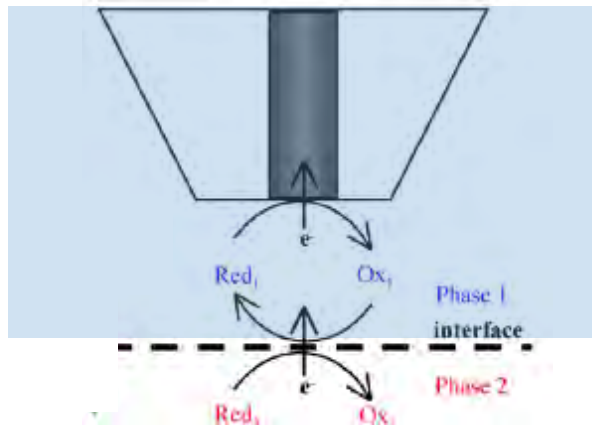
larger than in air (~ 2)
due to the hydration
of the polar head

Electrical Scanning Probe Microscopies

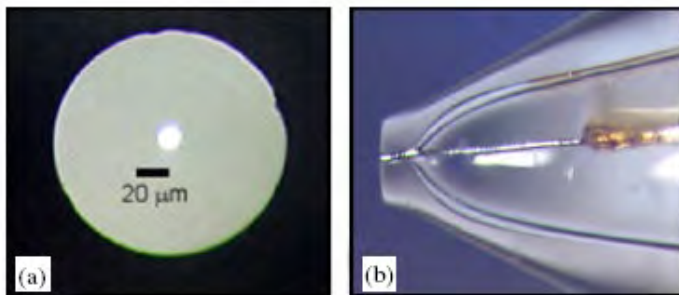


Scanning ElectroChemical Microscopy (SECM)

A scanning ultra-micro-electrode probe in solution, either amperometric or potentiometric, to investigate the **local electrochemical activity** (oxidation-reduction) of an interface.



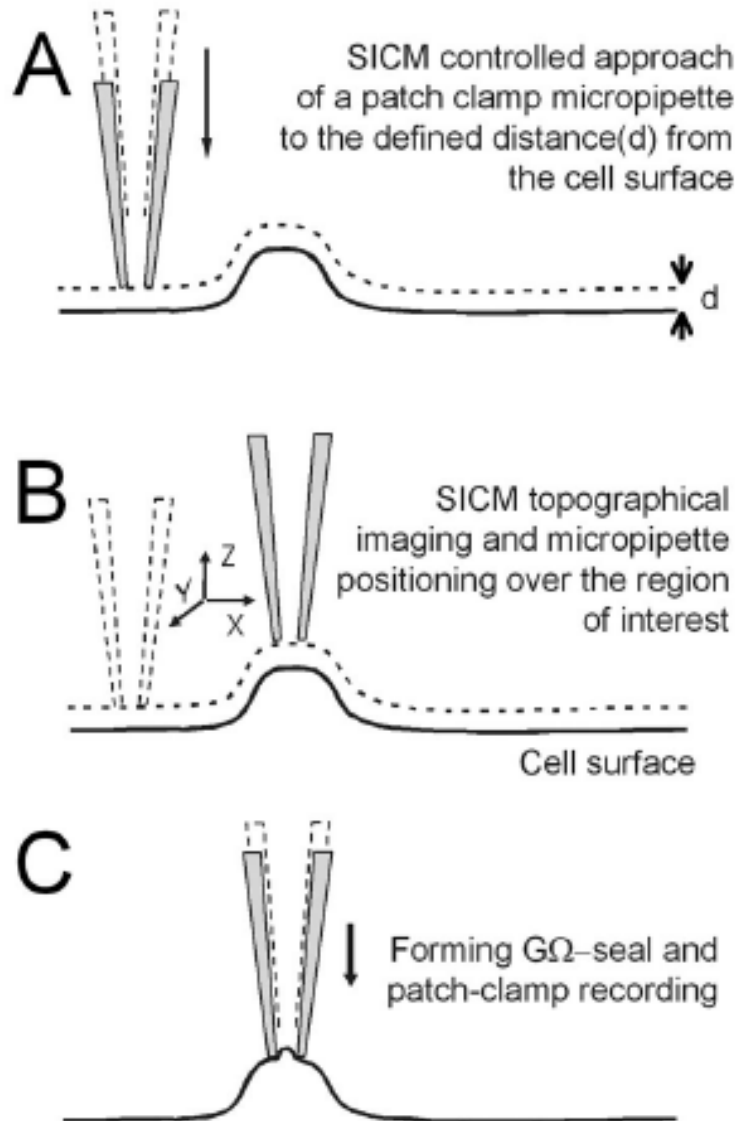
The ultra-micro-electrode probe (UME)



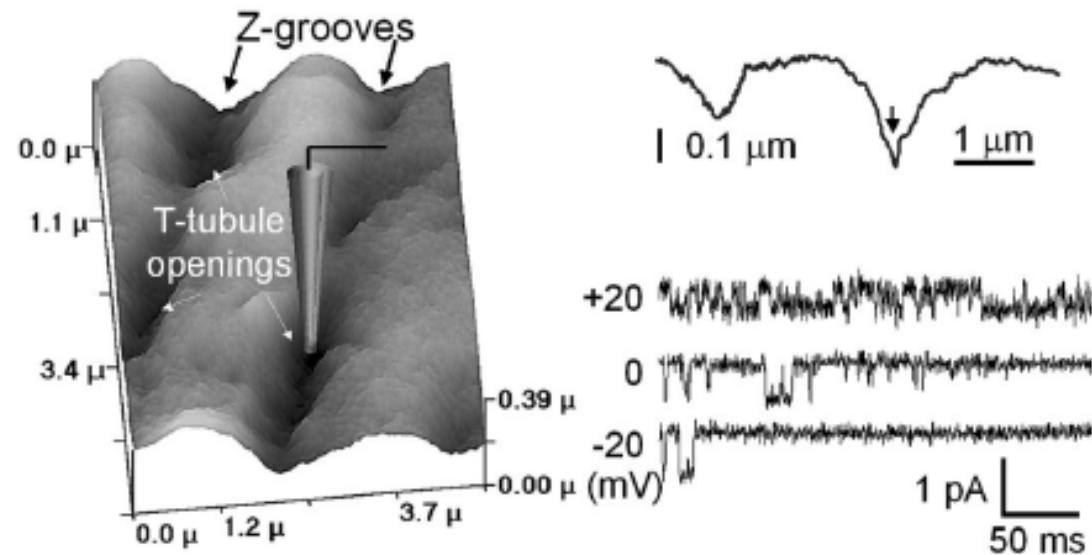
a disc-shaped electrode (a diameter of 0.6–25 μm)
with a sealed conductive wire in a glass capillary

Kwak J and Bard A. J. *Anal. Chem.* 1989
Edwards M. A. *et al* *Physiol. Meas.* 2006

Scanning Ion Conductance Microscopy (SICM)

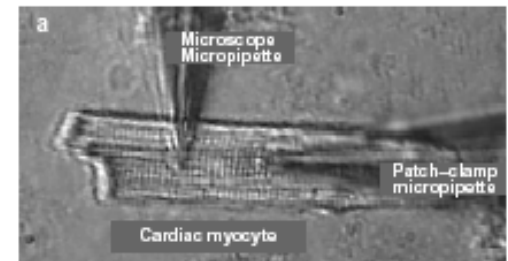
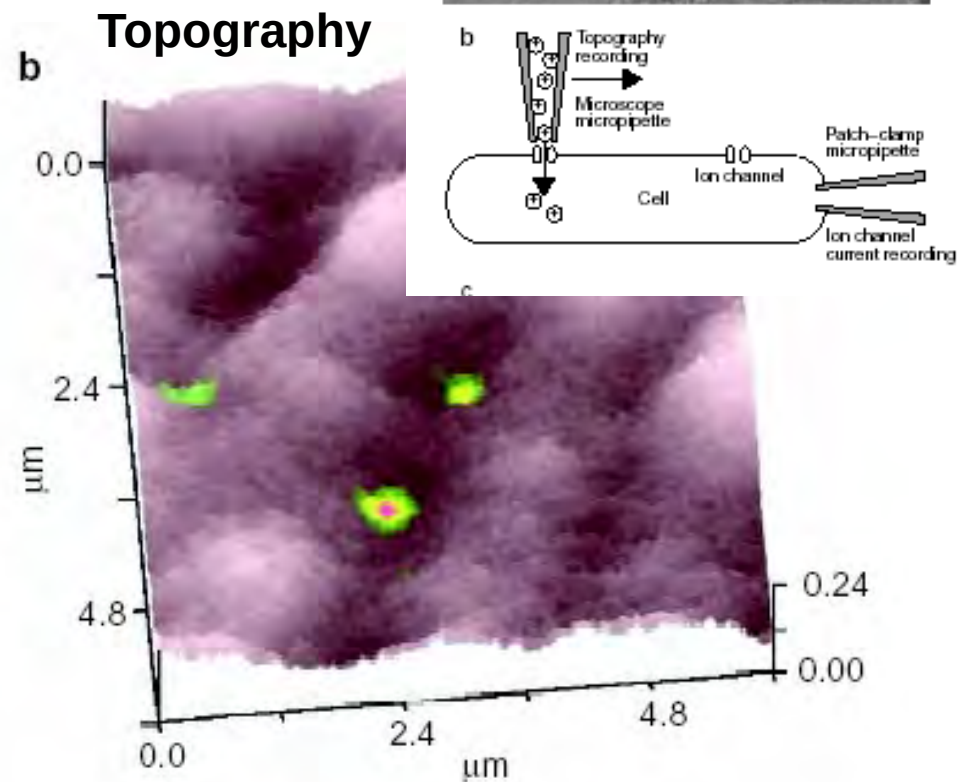
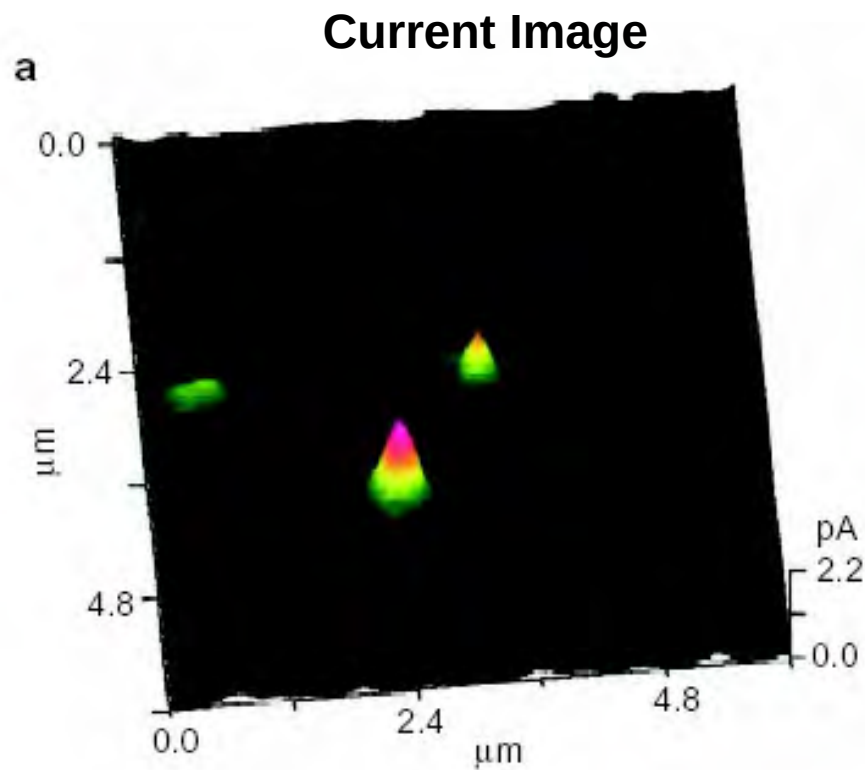


**Patch-clamp
with nanoscale positioning**



Scanning Ion Conductance Microscopy (SICM)

Detecting ion channels in living cells



Questions?



Dept. of Physics & Astronomy
Condensed Matter Physics

**PhD students and Postdoctoral
positions available**
laura.fumagalli@manchester.ac.uk